

## Appendix A: Substance-independent input data

Parameter	Unit	Value
<b>ENVIRONMENTAL DISTRIBUTION</b>		
Density of the solid phase	$\text{kg}\cdot\text{m}^{-3}$	2500
Density of water	$\text{kg}\cdot\text{m}^{-3}$	1000
Density of air	$\text{kg}\cdot\text{m}^{-3}$	1.3
Temperature at the continental air-water interface	K	285
Temperature at the moderate air-water interface	K	285
Temperature at the arctic air-water interface	K	263
Temperature at the tropic air-water interface	K	298
Constant of Junge equation	$\text{Pa}\cdot\text{m}$	<sup>a</sup>
Surface area of aerosol particles	$\text{m}^2\cdot\text{m}^{-3}$	<sup>a</sup>
Gas constant	$\text{Pa}\cdot\text{m}^3\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$	8.314
Volume fraction of solids in suspended matter	$\text{m}_{\text{solid}}^{-3}\cdot\text{m}_{\text{susp}}^{-3}$	0.1
Volume fraction of water in suspended matter	$\text{m}_{\text{water}}^{-3}\cdot\text{m}_{\text{susp}}^{-3}$	0.9
Weight fraction of organic carbon in suspended solids	$\text{kg}_{\text{oc}}\cdot\text{kg}_{\text{solid}}^{-1}$	0.1
Volume fraction of solids in sediment	$\text{m}_{\text{solid}}^{-3}\cdot\text{m}_{\text{sed}}^{-3}$	0.2
Volume fraction of water in sediment	$\text{m}_{\text{water}}^{-3}\cdot\text{m}_{\text{sed}}^{-3}$	0.8
Weight fraction of organic carbon in sediment solids	$\text{kg}_{\text{oc}}\cdot\text{kg}_{\text{solid}}^{-1}$	0.05
Volume fraction of solids in soil	$\text{m}_{\text{solid}}^{-3}\cdot\text{m}_{\text{soil}}^{-3}$	0.6
Volume fraction of water in soil	$\text{m}_{\text{water}}^{-3}\cdot\text{m}_{\text{soil}}^{-3}$	0.2
Volume fraction of air in soil	$\text{m}_{\text{water}}^{-3}\cdot\text{m}_{\text{soil}}^{-3}$	0.2
Weight fraction of organic carbon in sediment solids	$\text{kg}_{\text{oc}}\cdot\text{kg}_{\text{solid}}^{-1}$	0.02
Concentration of OH-radicals in atmosphere	$\text{molecules}\cdot\text{m}^{-3}$	$5\cdot10^{11}$
Fraction of sediment compartment that is aerated	$\text{m}^3\cdot\text{m}^{-3}$	0.1
Atmospheric mixing height at all scales	m	1000
Wind speed at all scales	$\text{m}\cdot\text{d}^{-1}$	$2.59\cdot10^5$
Aerosol-deposition velocity at all scales	$\text{m}\cdot\text{d}^{-1}$	86.4
Aerosol-collection efficiency at all scales	-	$2\cdot10^5$
Average daily precipitation at continental scale and moderate zone	$\text{mm}\cdot\text{yr}^{-1}$	700
Average daily precipitation in arctic zone	$\text{mm}\cdot\text{yr}^{-1}$	250
Average daily precipitation in tropic zone	$\text{mm}\cdot\text{yr}^{-1}$	1300
Concentration of biota in water at all scales	$\text{kg}_{\text{wwt}}\cdot\text{m}^{-3}$	0.1
Total area of the continental scale	$\text{km}^2$	$7.16\cdot10^6$
Total area of the moderate scale <sup>b</sup>	$\text{km}^2$	$4.25\cdot10^7$
Total area of the arctic scale	$\text{km}^2$	$7.78\cdot10^7$
Total area of the tropic scale	$\text{km}^2$	$1.28\cdot10^8$
Fraction of the continent that is fresh water	-	0.015
Fraction of the continent that is seawater	-	0.5
Fraction of the continent that is natural soil	-	0.3
Fraction of the continent that is agricultural soil	-	0.135
Fraction of the continent that is industrial soil	-	0.05
Fraction of the moderate zone that is water	-	0.5
Fraction of the moderate zone that is soil	-	0.5
Fraction of the arctic zone that is water	-	0.6
Fraction of the arctic zone that is soil	-	0.4
Fraction of the tropic zone that is water	-	0.7
Fraction of the tropic zone that is soil	-	0.3
Water depth of fresh water at continental scale	m	3
Water depth of sea water at continental scale	m	200
Water depth of sea water at global zones	m	1000
Suspended solids conc. continental fresh water	$\text{kg}_{\text{dwt}}\cdot\text{m}^{-3}$	0.025
Suspended solids conc. in sea water at all scales	$\text{kg}_{\text{dwt}}\cdot\text{m}^{-3}$	0.005
Sediment mixing depth at all scales	m	0.03
Settling velocity of suspended particles at all scales	$\text{m}\cdot\text{d}^{-1}$	2.5
Production rate of susp. matter in continental fresh water	$\text{kg}_{\text{dwt}}\cdot\text{m}^{-2}\cdot\text{yr}^{-1}$	0.01
Production rate of susp. matter in seawater at all scales	$\text{kg}_{\text{dwt}}\cdot\text{m}^{-2}\cdot\text{yr}^{-1}$	0.001
Fraction of rain water that infiltrates the soil at all scales	-	0.25
Fraction of wet prec. that runs off soil to water at all scales	-	0.25
Soil-erosion rate at all scales	$\text{m}\cdot\text{d}^{-1}$	$8.2\cdot10^{-8}$
Enthalpy of vaporation	$\text{kJ}\cdot\text{mol}^{-1}$	50
Enthalpy of solution	$\text{kJ}\cdot\text{mol}^{-1}$	10
Air-film partial mass-transfer coefficient at all scales (air-water and air-soil interfaces)	$\text{m}\cdot\text{d}^{-1}$	120
Water-film partial mass-transfer coefficient at all scales (air-water interface)	$\text{m}\cdot\text{d}^{-1}$	1.2
Soil-air partial mass-transfer coefficient at all scales (air-soil interface)	$\text{m}\cdot\text{d}^{-1}$	0.48
Soil water-water film partial-mass transfer coefficient (air-soil interface)	$\text{m}\cdot\text{d}^{-1}$	$4.8\cdot10^{-5}$

Water-film partial mass-transfer coefficient at all scales (sediment-water interface)	$\text{m} \cdot \text{d}^{-1}$	0.24
Pore water partial mass-transfer coefficient (sediment-water interface)	$\text{m} \cdot \text{d}^{-1}$	$2.4 \cdot 10^{-3}$
<sup>a</sup> By default, the product of CONjunge and SURFaer is set to $1 \cdot 10^{-4}$ Pa, although CONjunge is (partly) substance-dependent; <sup>b</sup> the area of the continental scale is not included in the area of the moderate scale.		

Parameter	Unit	Value
<b>HUMAN EXPOSURE ASSESSMENT</b>		
Volume fraction of fat in plant roots	$\text{m}^3 \cdot \text{m}^{-3}$	0.005
Volume fraction of water in plant roots	$\text{m}^3 \cdot \text{m}^{-3}$	0.93
Bulk density of plant root tissue	$\text{kg}_{\text{wwt}} \cdot \text{m}^{-3}$	1000
Volume fraction of fat in plant leafs	$\text{m}^3 \cdot \text{m}^{-3}$	0.01
Volume fraction of water in plant leafs	$\text{m}^3 \cdot \text{m}^{-3}$	0.65
Volume fraction of air in plant leafs	$\text{m}^3 \cdot \text{m}^{-3}$	0.3
Bulk density of plant leaf tissue	$\text{kg}_{\text{wwt}} \cdot \text{m}^{-3}$	800
Leaf surface area	$\text{m}^2$	5
Shoot volume	$\text{m}^3$	0.002
Transpiration stream	$\text{m}^3 \cdot \text{d}^{-1}$	0.001
Correction factor for differences between plant lipids and octanol	-	0.95
Growth-rate constant for dilution by growth	$\text{d}^{-1}$	0.035
Daily intake of grass by cattle	$\text{kg}_{\text{dwt}} \cdot \text{d}^{-1}$	72.5
Daily intake of soil by cattle	$\text{kg}_{\text{dwt}} \cdot \text{d}^{-1}$	0.4
Daily intake of air by cattle	$\text{m}^3 \cdot \text{d}^{-1}$	122
Daily intake of drinking water by cattle	$\text{m}^3 \cdot \text{d}^{-1}$	0.055
Conversion dry weight to wet weight grass	$\text{kg}_{\text{wwt}} \cdot \text{kg}_{\text{dwt}}^{-1}$	4
Fraction drinking water from groundwater	-	0.57
Fraction drinking water from surface water	-	0.43
Daily intake of drinking water	$\text{l} \cdot \text{d}^{-1}$	1.14
Daily intake of fish <sup>a</sup>	$\text{kg}_{\text{wwt}} \cdot \text{d}^{-1}$	0.03
Daily intake of leaf crops (incl. fruit and cereals) <sup>b</sup>	$\text{kg}_{\text{wwt}} \cdot \text{d}^{-1}$	0.77
Daily intake of root crops <sup>b</sup>	$\text{kg}_{\text{wwt}} \cdot \text{d}^{-1}$	0.18
Daily intake of meat	$\text{kg}_{\text{wwt}} \cdot \text{d}^{-1}$	0.26
Daily intake of dairy products	$\text{kg}_{\text{wwt}} \cdot \text{d}^{-1}$	0.28
Daily inhalation rate	$\text{m}^3 \cdot \text{d}^{-1}$	12.7
Body weight	$\text{kg}$	70
Daily soil ingestion <sup>c</sup>	$\text{mg}_{\text{wwt}} \cdot \text{d}^{-1}$	50

<sup>a</sup> it is assumed that 90% of the total fish intake ( $\text{kg}_{\text{wwt}}$ ) on the continental scale are salt water species and 10% fresh water species; <sup>b</sup> an edible fraction of 0.8 for fruit, vegetables and root crops is assumed; <sup>c</sup> it is assumed that on the continental scale all ingested soil comes from industrial/urban soils.

## Appendix B: Substance-specific input data

Table B.1a: Substance-specific input parameters for metals

Compound name CAS nr.	Unit	Antimony 7440-36-0	Arsenic 7440-38-2	Barium 7440-39-3	Beryllium 7440-41-7	Cadmium 7440-43-9	Chromium III 7440-47-3	ChromiumVI 7440-47-3	Cobalt 7440-48-4	Copper 7440-50-8	Source
<b>Effects assessment</b>											
Oral Human Limit Value	$\text{kg} \cdot \text{kg}(\text{bw})^{-1} \cdot \text{d}^{-1}$	$8.6 \cdot 10^{-10}$	$2.14 \cdot 10^{-9}$	$7.0 \cdot 10^{-8}$	$5.0 \cdot 10^{-10}$	$1.0 \cdot 10^{-9}$	$5.0 \cdot 10^{-9}$	$3.0 \cdot 10^{-9}$	$1.4 \cdot 10^{-9}$	$1.4 \cdot 10^{-7}$	1-9
Inhalatory Human Limit Value	$\text{kg} \cdot \text{m}^{-3}$	$3.2 \cdot 10^{-9}$	$2.5 \cdot 10^{-13}$	$5.0 \cdot 10^{-10}$	$4.0 \cdot 10^{-13}$	$6.0 \cdot 10^{-13}$		$2.5 \cdot 10^{-14}$	$5.0 \cdot 10^{-12}$	$2.0 \cdot 10^{-11}$	
Aquatic Predicted No Effect Concentration	$\text{kg} \cdot \text{m}^{-3}$	$4.6 \cdot 10^{-4}$	$2.4 \cdot 10^{-5}$	$5.8 \cdot 10^{-5}$	$1.6 \cdot 10^{-7}$	$3.4 \cdot 10^{-7}$	$3.4 \cdot 10^{-5}$	$8.5 \cdot 10^{-6}$	$2.6 \cdot 10^{-6}$	$1.1 \cdot 10^{-6}$	
Terrestrial Predicted No Effect Concentration	$\text{kg} \cdot \text{kg}(\text{dwt})^{-1}$	EP <sup>a</sup>	$3.6 \cdot 10^{-7}$	EP <sup>a</sup>	EP <sup>a</sup>	$4.6 \cdot 10^{-7}$	$3.3 \cdot 10^{-7}$	$3.3 \cdot 10^{-7}$	EP <sup>a</sup>	$7.7 \cdot 10^{-6}$	
<b>Physico-chemical properties</b>											
molecular weight	$\text{g} \cdot \text{mol}^{-1}$	121.76	74.92	137.33	9.01	112.41	52	52	5893	63.55	10
melting point	$^{\circ}\text{C}$	630.74	817	725	1278	320.9	1900	1900	1495	1083	
<b>Partition coefficients</b>											
solid-water partition coefficient soil	$\text{l} \cdot \text{kg}^{-1}$	85	3337	60	38	152	8427	8427	40	221	11, 12
solid-water partition coefficient sediment	$\text{l} \cdot \text{kg}^{-1}$	2570	6607	1000	603	85114	190546	190546	3981	33884	
solid-water partition coefficient suspended matter	$\text{l} \cdot \text{kg}^{-1}$	3715	10000	1349	851	128825	288403	288403	3890	50119	
aerosol collection efficiency		$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	$1.0 \cdot 10^{-5}$	
fraction of aerosol bounded substance	-	0.95	0.95	0.95	0.95	0.95	0.95	0.95	0.95	0.95	
<b>Exposure assessment</b>											
bioconcentration factor in fish	$\text{l} \cdot \text{kg}(\text{wwt})^{-1}$	80	150	216	19	38	40	40	1	120	13-23
bioconcentration factor from soil to plant roots	$\text{kg}(\text{wwt}) \cdot \text{kg}(\text{wwt})^{-1}$	$2.0 \cdot 10^{-2}$	$1.6 \cdot 10^{-2}$	$2.0 \cdot 10^{-2}$	$6.1 \cdot 10^{-3}$	$1.4 \cdot 10^{-1}$	$4.0 \cdot 10^{-3}$	$4.0 \cdot 10^{-3}$	$6.1 \cdot 10^{-3}$	$2.0 \cdot 10^{-2}$	
bioconcentration factor from soil to plant leafs	$\text{kg}(\text{wwt}) \cdot \text{kg}(\text{wwt})^{-1}$	$1.2 \cdot 10^{-2}$	$9.4 \cdot 10^{-3}$	$5.9 \cdot 10^{-4}$	$3.5 \cdot 10^{-3}$	$1.8 \cdot 10^{-2}$	$2.3 \cdot 10^{-4}$	$2.3 \cdot 10^{-4}$	$1.8 \cdot 10^{-3}$	$1.2 \cdot 10^{-2}$	
biotransfer factor for meat	$\text{d} \cdot \text{kg}(\text{food})^{-1}$	$3.9 \cdot 10^{-4}$	$2.2 \cdot 10^{-4}$	$9.7 \cdot 10^{-5}$	$3.2 \cdot 10^{-6}$	$5.3 \cdot 10^{-3}$	$9.2 \cdot 10^{-3}$	$9.2 \cdot 10^{-3}$	$9.7 \cdot 10^{-3}$	$1.3 \cdot 10^{-2}$	
biotransfer factor for milk	$\text{d} \cdot \text{kg}(\text{food})^{-1}$	$1.1 \cdot 10^{-4}$	$6.0 \cdot 10^{-5}$	$3.4 \cdot 10^{-4}$	$8.8 \cdot 10^{-7}$	$1.5 \cdot 10^{-3}$	$1.1 \cdot 10^{-3}$	$1.1 \cdot 10^{-3}$	$2.8 \cdot 10^{-3}$	$1.7 \cdot 10^{-3}$	
bioavailability for human inhalation	-		0.4	0.75	0.5	0.15	0.1	0.25			
bioavailability for human oral uptake	-	0.1	0.9	0.1	0.001	0.05	0.01	0.05		0.4	

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> RIZA (1999); <sup>2</sup> Vermeire et al. (1991); <sup>3</sup> Huijbregts (1999); <sup>4</sup> USEPA (1998b); <sup>5</sup> WHO (1987a); <sup>6</sup> JECFA (1989); <sup>7</sup> Environmental Defense Fund (1999); <sup>8</sup> Janssen et al. (1995); <sup>9</sup> Janus et al. (1994); <sup>10</sup> Cambridgesoft Corporation (1998); <sup>11</sup> Crommentuijn et al. (1997a); <sup>12</sup> De Groot et al. (1998); <sup>13</sup> Slooff et al. (1992c); <sup>14</sup> Owen (1990); <sup>15</sup> Spectrum Research (1992); <sup>16</sup> Van de Plassche (1994); <sup>17</sup> Slooff et al. (1990b); <sup>18</sup> Van de Berg (1995); <sup>19</sup> Ng (1982); <sup>20</sup> Slooff et al. (1992b); <sup>21</sup> Owen (1990); <sup>22</sup> Slooff et al. (1990a); <sup>23</sup> personal assessment

Table B.1b: Substance-specific input parameters for metals

Compound name	Unit	Lead	Mercury	Methyl-mercury	Molybdenum	Nickel	Selenium	Thallium	Tin	Vanadium	Zinc	Source
CAS nr.	-	7439-92-1	7439-97-6	22967-92-6	7439-98-7	7440-02-0	7782-49-2	7440-28-0	7440-31-5	7440-62-2	7440-66-6	
<b>Effects assessment</b>												1-11
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.6·10 <sup>-9</sup>	7.2·10 <sup>-10</sup>	4.71·10 <sup>-10</sup>	1.0·10 <sup>-8</sup>	5.0·10 <sup>-9</sup>	5.0·10 <sup>-9</sup>	2.0·10 <sup>-10</sup>	2.0·10 <sup>-6</sup>	2.0·10 <sup>-9</sup>	1.0·10 <sup>-6</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>		3.0·10 <sup>-10</sup>	1.0·10 <sup>-9</sup>			2.5·10 <sup>-12</sup>	8.0·10 <sup>-11</sup>		1.0·10 <sup>-9</sup>	9.0·10 <sup>-10</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.1·10 <sup>-5</sup>	1.3·10 <sup>-7</sup>	1.0·10 <sup>-8</sup>	2.9·10 <sup>-5</sup>	1.8·10 <sup>-6</sup>	5.3·10 <sup>-6</sup>	1.6·10 <sup>-6</sup>	1.8·10 <sup>-5</sup>	8.2·10 <sup>-7</sup>	6.6·10 <sup>-6</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	4.4·10 <sup>-5</sup>	1.5·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	7.4·10 <sup>-7</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	6.7·10 <sup>-6</sup>	
<b>Physico-chemical properties</b>												12, 13
molecular weight	g.mol <sup>-1</sup>	207.2	200.59	215.62	95.94	58.69	78.96	204.38	118.69	50.94	65.39	
melting point	°C	327.43	-38.9	-38.9	2622	1455	217	302	231.9	1344	419.5	
<b>Partition coefficients</b>												14, 15
solid-water partition coefficient soil	1.kg <sup>-1</sup>	4332	170	170	871	359	20	334	4332	309	334	
solid-water partition coefficient sediment	1.kg <sup>-1</sup>	4.3·10 <sup>5</sup>	1.1·10 <sup>5</sup>	112201.8	851	5.2·10 <sup>3</sup>	417	1.0·10 <sup>3</sup>	1.2·10 <sup>6</sup>	3.9·10 <sup>3</sup>	7.2·10 <sup>4</sup>	
solid-water partition coefficient suspended matter	1.kg <sup>-1</sup>	6.5·10 <sup>5</sup>	1.7·10 <sup>5</sup>	169824.4	1.1·10 <sup>3</sup>	7.9·10 <sup>3</sup>	589	1.5·10 <sup>3</sup>	3.7·10 <sup>5</sup>	5.5·10 <sup>3</sup>	1.1·10 <sup>5</sup>	
aerosol collection efficiency	-	1.0·10 <sup>5</sup>	2.0·10 <sup>5</sup>	2.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	1.0·10 <sup>5</sup>	
fraction of aerosol bounded substance	-	0.95	0.05	0.05	0.95	0.95	0.95	0.95	0.95	0.95	0.95	
<b>Exposure assessment</b>												5, 16-29
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>	500	3030	21700	216	87	500	250	216	200	1000	
bioconcentration factor from soil to plant roots	kg(wwt).kg(wwt) <sup>-1</sup>	6.1·10 <sup>-3</sup>	6.1·10 <sup>-3</sup>	6.1·10 <sup>-3</sup>	6.1·10 <sup>-2</sup>	2.0·10 <sup>-2</sup>	2.0·10 <sup>-2</sup>	6.1·10 <sup>-3</sup>	6.1·10 <sup>-3</sup>	6.1·10 <sup>-2</sup>	8.1·10 <sup>-2</sup>	
bioconcentration factor from soil to plant leafs	kg(wwt).kg(wwt) <sup>-1</sup>	1.2·10 <sup>-4</sup>	1.8·10 <sup>-3</sup>	1.8·10 <sup>-3</sup>	1.8·10 <sup>-3</sup>	8.2·10 <sup>-3</sup>	1.2·10 <sup>-2</sup>	3.5·10 <sup>-1</sup>	1.8·10 <sup>-3</sup>	3.5·10 <sup>-1</sup>	1.2·10 <sup>-2</sup>	
biotransfer factor for meat	d.kg(food) <sup>-1</sup>	1.0·10 <sup>-3</sup>	1.7·10 <sup>-3</sup>	1.2·10 <sup>-2</sup>	6.8·10 <sup>-3</sup>	2.0·10 <sup>-3</sup>	1.4·10 <sup>-2</sup>	6.7·10 <sup>-3</sup>	4.2·10 <sup>-3</sup>	6.7·10 <sup>-4</sup>	1.2·10 <sup>-1</sup>	
biotransfer factor for milk	d.kg(food) <sup>-1</sup>	2.5·10 <sup>-4</sup>	4.6·10 <sup>-4</sup>	3.3·10 <sup>-3</sup>	1.4·10 <sup>-3</sup>	9.7·10 <sup>-4</sup>	3.9·10 <sup>-3</sup>	1.8·10 <sup>-3</sup>	1.2·10 <sup>-3</sup>	1.8·10 <sup>-4</sup>	9.7·10 <sup>-3</sup>	
respirable fraction of inhaled substance	-								0.35			
bioavailability for human inhalation	-	0.5	0.75	1.0		0.06	0.3	1		0.25	0.5	
bioavailability for human oral uptake	-	0.1	0.0001	0.95		0.05	0.6		0.05	0.02	0.5	

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> RIZA (1999); <sup>2</sup> USEPA (1998b); <sup>3</sup> WHO (1987a); <sup>4</sup> Vermeire et al. (1991); <sup>5</sup> Janssen et al. (1998); <sup>6</sup> JECFA (1982); <sup>7</sup> JECFA (1986); <sup>8</sup> JECFA (1989); <sup>9</sup> Environmental Defense Fund (1999); <sup>10</sup> Huijbregts (1999); <sup>11</sup> Janus et al. (1994); <sup>12</sup> Cambridgesoft Corporation (1998); <sup>13</sup> Lide (1993); <sup>14</sup> Crommentuijn et al. (1997a); <sup>15</sup> De Groot et al. (1998); <sup>16</sup> Guinée et al. (1996a); <sup>17</sup> Bockting et al. (1996); <sup>18</sup> Slooff et al. (1995); <sup>19</sup> Van de Plassche et al. (1992); <sup>20</sup> Slooff et al. (1992a); <sup>21</sup> Slooff et al. (1993); <sup>22</sup> Owen (1990); <sup>23</sup> Van de Berg (1995); <sup>24</sup> Ng (1982); <sup>25</sup> WHO (1995a); <sup>26</sup> WHO (1996a); <sup>27</sup> Slooff et al. (1993); <sup>28</sup> Spectrum Research (1992); <sup>29</sup> personal assessment

Table B.2: Substance-specific input parameters for inorganics

Compound name	Unit	ammonia (anhydrous)	nitrogen dioxide	sulphur dioxide	hydrogen sulphide	hydrogen chloride	primary fine particulate matter (PM10)	Source
CAS nr.	-	7664-41-7	10102-44-0	7446-09-5	7783-06-4	7647-01-0		
<b>Effects assessment</b>								1-3
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	1·10 <sup>-7</sup>	4·10 <sup>-8</sup>	3.5·10 <sup>-7</sup>	1.5·10 <sup>-7</sup>	2·10 <sup>-8</sup>	4·10 <sup>-8</sup>	
<b>Physico-chemical properties</b>								2, 5-9
molecular weight	g.mol <sup>-1</sup>	17.03	46.02	64.06	34.08	36.461		
melting point	°C	-77.7	-11.2	-75.51	-85.6	-114.24		
vapor pressure (25 °C)	Pa	9.2·10 <sup>5</sup>	1.4·10 <sup>5</sup>	3.1·10 <sup>5</sup>	2.0·10 <sup>6</sup>	1.4·10 <sup>5</sup>	1·10 <sup>-30</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	5.7·10 <sup>5</sup>	1.0·10 <sup>6</sup>	1.1·10 <sup>5</sup>	2.9·10 <sup>3</sup>	6.2·10 <sup>5</sup>	1·10 <sup>-13</sup>	
<b>Partition coefficients</b>								7, 10-12
Henry's law constant (25 °C)	Pa.m <sup>3</sup> .mol <sup>-1</sup>				5.5·10 <sup>7</sup>			
aerosol deposition velocity	m.s <sup>-1</sup>						1.5·10 <sup>-3</sup>	
scavenging ratio	-	1.4·10 <sup>6</sup>		3·10 <sup>5</sup>	<sup>a</sup> 3·10 <sup>5</sup>	<sup>a</sup> 1.4·10 <sup>6</sup>	3.5·10 <sup>5</sup>	
fraction of aerosol bounded substance	-						1	
<b>Degradation rates</b>								12-14
reaction half-life in air	d	1·10 <sup>30</sup>		3.6			1·10 <sup>30</sup>	
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>		1.1·10 <sup>-11</sup>		4.8·10 <sup>-12</sup>	8.1·10 <sup>-13</sup>		

<sup>a</sup>It is assumed that HCl has a scavenging ratio equal to HNO<sub>3</sub> and H<sub>2</sub>S has a scavenging ratio equal to SO<sub>2</sub> (values for HNO<sub>3</sub> and SO<sub>2</sub> taken from Barrett & Berge, 1996); <sup>1</sup> WHO-EURO (1987); <sup>2</sup> WHO (1997a); <sup>3</sup> USEPA (1998b); <sup>4</sup> Guinée et al. (1996a); <sup>5</sup> Verschueren (1996); <sup>6</sup> Cambridgesoft Corporation (1998); <sup>7</sup> Slooff et al. (1991a); <sup>8</sup> University of Akron (1998); <sup>9</sup> Howard Hughes Medical Institute (1998); <sup>10</sup> D. van de Meent (pers. comm.); <sup>11</sup> F. de Leeuw (pers. comm.); <sup>12</sup> Barrett & Berge (1996); <sup>13</sup> Lide (1993); <sup>14</sup> personal assessment

Table B.3: Substance-specific input parameters for non-aromatics

Compound name	Unit	acrylonitrile	acrolein	1,3-butadiene	carbon disulfide	ethylene	formaldehyde	propylene oxide	Source
CAS nr.	-	107-13-1	107-02-8	106-99-0	75-15-0	74-85-1	50-00-0	75-56-9	
<b>Effects assessment</b>									1-8
oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>1</sup>	1.0·10 <sup>-12</sup>	5.0·10 <sup>-9</sup>				1.5·10 <sup>-7</sup>	2.9·10 <sup>-12</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	5.0·10 <sup>-11</sup>	5.0·10 <sup>-10</sup>	4.0·10 <sup>-12</sup>	1.0·10 <sup>-7</sup>	1.0·10 <sup>-7</sup>	1.0·10 <sup>-7</sup>	9.0·10 <sup>-10</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	7.6·10 <sup>-6</sup>	7.0·10 <sup>-9</sup>	7.1·10 <sup>-5</sup>	2.1·10 <sup>-6</sup>	8.5·10 <sup>-3</sup>	2.1·10 <sup>-6</sup>	1.7·10 <sup>-4</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>									9-15
molecular weight	g.mol <sup>-1</sup>	53.06	56.04	54.09	76.12	28.054	30.03	58.08	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	1.8	0.1	97.7	87	13.5	2.2	1.1	
melting point	°C	-83.55	-87	-108.9	-111.2	-169.14	-92	-112.13	
vapor pressure (25 °C)	Pa	1.1·10 <sup>4</sup>	3.65·10 <sup>4</sup>	2.8·10 <sup>5</sup>	4.9·10 <sup>4</sup>	2.7·10 <sup>9</sup>	4.5·10 <sup>5</sup>	7.1·10 <sup>4</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	7.55·10 <sup>4</sup>	2.08·10 <sup>5</sup>	735	2400	260	1.22·10 <sup>6</sup>	4.76·10 <sup>5</sup>	
<b>Partition coefficients</b>									9, 10, 16, 17
Henry's law constant (25 °C)	Pa.m <sup>-3</sup> .mol <sup>1</sup>	11.1	2.4	2.57·10 <sup>5</sup>			3.2·10 <sup>-2</sup>	8.5	
organic carbon partition coefficient	l.kg <sup>-1</sup>		0.5	129					13, 18-23
<b>Degradation rates</b>									
reaction half-life in air	d	4.95			9		0.15	30.9	
hydroxyl radical reaction in air (-10 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>		2.2·10 <sup>-11</sup>	8.2·10 <sup>-11</sup>		1.0·10 <sup>-11</sup>			
hydroxyl radical reaction in air (12 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>		2.0·10 <sup>-11</sup>	7.2·10 <sup>-11</sup>		9.1·10 <sup>-12</sup>			
hydroxyl radical reaction in air (25 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>		1.9·10 <sup>-11</sup>	6.7·10 <sup>-11</sup>		8.5·10 <sup>-12</sup>			
hydrolysis in water, soil, sediment (pH=6; 12 °C)	d								13
hydrolysis in water, soil, sediment (pH=7; 12 °C)	d								14
hydrolysis in water, soil, sediment (pH=8; 12 °C)	d								14
biodegradation in surface water (12C)	d	9.4	24.4	24.4	30	9.2	4.6	30	
biodegradation in soil (12 °C)	d	9.4	24.4	24.4	100	9.2	4.6	100	
aerobic biodegradation in the sediment zone (12 °C)	d	9.4	24.4	24.4	100	9.2	4.6	100	
anaerobic biodegradation in the sediment zone (12 °C)	d	37.4	101	97	400	36.9	18.5	400	
<b>Exposure assessment</b>									19, 24-27
bioconcentration factor in fish	l.kg(wwt) <sup>1</sup>		344						
bioavailability for inhalation	-	0.95	0.82	0.65			1	1	
bioavailability for oral uptake	-	0.98							

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> WHO-EURO (1987); <sup>2</sup> WHO-EURO (1996); <sup>3</sup> Janus et al. (1994); <sup>4</sup> RIZA (1999); <sup>5</sup> Environmental Defense Fund (1999); <sup>6</sup> Huijbregts (1999); <sup>7</sup> USEPA (1998b); <sup>8</sup> Janssen et al. (1995); <sup>9</sup> Mackay et al. (1993); <sup>10</sup> Mackay et al. (1995); <sup>11</sup> Mackay et al. (1997); <sup>12</sup> Cambridgesoft Corporation (1998); <sup>13</sup> Howard (1989); <sup>14</sup> Environmental Science Center (1998); <sup>15</sup> Verschueren (1996); <sup>16</sup> Lide (1993); <sup>17</sup> Bockting et al. (1993); <sup>18</sup> Howard et al. (1991); <sup>19</sup> Slooff (1988); <sup>20</sup> Jager et al. (1997); <sup>21</sup> RIVM et al. (1998); <sup>22</sup> Howard (1990); <sup>23</sup> Atkinson (1986); <sup>24</sup> WHO (1989c); <sup>25</sup> Slooff et al. (1991c); <sup>26</sup> WHO (1992); <sup>27</sup> Owen (1990)

Table B.4a: Substance-specific input parameters for aromatics

Compound name	Unit	Benzene	Toluene	Styrene	Phenol	Ethylbenzene	m-xylene	o-xylene	p-xylene	Butylbenzyl-phthalate	Source
CAS nr.	-	71-43-2	108-88-3	100-42-5	108-95-2	100-41-4	108-38-3	95-47-6	106-42-3	85-68-7	
<b>Effects assessment</b>											1-8
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	1.0·10 <sup>-7</sup>	2.0·10 <sup>-7</sup>	7.7·10 <sup>-8</sup>	6.0·10 <sup>-8</sup>	1.36·10 <sup>-7</sup>	1.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	2.0·10 <sup>-7</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	1.67·10 <sup>-10</sup>	2.6·10 <sup>-7</sup>	2.6·10 <sup>-7</sup>	6.0·10 <sup>-7</sup>	7.7·10 <sup>-8</sup>	1.0·10 <sup>-6</sup>	3.4·10 <sup>-7</sup>	1.0·10 <sup>-6</sup>	5.0·10 <sup>-8</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.4·10 <sup>-3</sup>	7.3·10 <sup>-4</sup>	5.7·10 <sup>-4</sup>	9.0·10 <sup>-7</sup>	3.7·10 <sup>-4</sup>	3.3·10 <sup>-4</sup>	4.0·10 <sup>-4</sup>	4.0·10 <sup>-4</sup>	7.5·10 <sup>-6</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	1.5·10 <sup>-7</sup>	EP <sup>a</sup>	5.6·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>											10-13
molecular weight	g.mol <sup>1</sup>	78.11	92.13	104.14	94.1	106.2	106.2	106.2	106.2	312.39	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	147	490	1122	29	1349	1585	1413	1514	56234	
melting point	°C	5.53	-95	-30.6	41	-95	-47.9	-25.2	13.2	-35	
vapor pressure (25 °C)	Pa	1.3·10 <sup>4</sup>	3.8·10 <sup>3</sup>	800	47	1.3·10 <sup>3</sup>	1.1·10 <sup>3</sup>	1.2·10 <sup>3</sup>	1.2·10 <sup>3</sup>	8.6·10 <sup>-4</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	1752	515	300	88360	152	160	220	215	2.7	
<b>Partition coefficients</b>											9, 10, 14-16
Henry's law constant (25 °C)	Pa.m <sup>3</sup> .mol <sup>1</sup>	5.9·10 <sup>2</sup>	9.2·10 <sup>2</sup>	2.8·10 <sup>2</sup>	4.0·10 <sup>-2</sup>	8.1·10 <sup>2</sup>	9.9·10 <sup>2</sup>	6.8·10 <sup>2</sup>	1.0·10 <sup>3</sup>		
organic carbon partition coefficient	l.kg <sup>-1</sup>	79	115	912	35	166	218	189	525	1622	
<b>Degradation rates</b>											17-19
reaction half-life in air	d			0.3	0.6	2.1		1.1	1.1	1.5	
hydroxyl radical reaction in air (-10 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	1.1·10 <sup>-12</sup>	7.1·10 <sup>-12</sup>				2.6·10 <sup>-11</sup>				
hydroxyl radical reaction in air (12 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	1.2·10 <sup>-12</sup>	6.5·10 <sup>-12</sup>				2.5·10 <sup>-11</sup>				
hydroxyl radical reaction in air (25 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	1.2·10 <sup>-12</sup>	6.2·10 <sup>-12</sup>				2.5·10 <sup>-11</sup>				
biodegradation in surface water (12 °C)	d	16	16	35	2	10	24	24	24	5	
biodegradation in soil (12 °C)	d	331	16	35	6	10	24	24	24	5	
aerobic biodegradation in sediment (12 °C)	d	16	16	35	2	10	24	24	24	5	
anaerobic biodegradation in sediment (12 °C)	d	495	189	138	26	349	98	444	98	124	
<b>Exposure assessment</b>											10-13, 20-26
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	6	8	14	17	16	19	17	19	12	
plant conductance	m.s <sup>-1</sup>				7.56·10 <sup>-4</sup>						
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>	2.9									
bioavailability for inhalation	-	0.47	0.5		0.75	0.64	0.64	0.64	0.64		
bioavailability for oral uptake	-	1	1		0.95	0.82	1	1	1	0.5	

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> WHO-EURO (1996); <sup>2</sup> Environmental Defense Fund (1999); <sup>3</sup> Rademaker et al. (1993); <sup>4</sup> Janus et al. (1994); <sup>5</sup> Vermeire et al. (1991); <sup>6</sup> USEPA (1998b); <sup>7</sup> RIZA (1999); <sup>8</sup> Huijbregts (1999); <sup>9</sup> Van de Plassche & Bockting (1993); <sup>10</sup> Mackay et al. (1992a); <sup>11</sup> Mackay et al. (1993); <sup>12</sup> Mackay et al. (1995); <sup>13</sup> Staples et al. (1997); <sup>14</sup> Lide (1993); <sup>15</sup> Sabljic et al. (1995); <sup>16</sup> Bockting et al. (1993); <sup>17</sup> Howard et al. (1991); <sup>18</sup> RIVM et al. (1998); <sup>19</sup> Atkinson (1986); <sup>20</sup> Nendza (1991); <sup>21</sup> Devillers et al. (1996); <sup>22</sup> Owen (1990); <sup>23</sup> WHO (1994b); <sup>24</sup> WHO (1996c); <sup>25</sup> Peijnenburg et al. (1991); <sup>26</sup> Riederer (1995)

Table B.4b: Substance-specific input parameters for aromatics

Compound name	Unit	Di(2-ethylhexyl)-phtalate	Dibutyl-phtalate	Diethyl-phtalate	Dihexyl-phtalate	Diisooctyl-phtalate	Diisodecyl-phtalate	Dimethyl-phtalate	Diocetyl-phtalate	Phtalic anhydride	Source
CAS nr.	-	117-81-7	84-74-2	84-66-2	84-75-3	27554-26-3	26761-40-0	133-11-3	117-84-0	85-44-9	
<b>Effects assessment</b>											1-7
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>1</sup>	2.5·10 <sup>-8</sup>	1.0·10 <sup>-7</sup>	8.0·10 <sup>-7</sup>	<sup>a</sup> 1.0·10 <sup>-9</sup>	1.0·10 <sup>-8</sup>	5.0·10 <sup>-8</sup>	<sup>a</sup> 1.0·10 <sup>-9</sup>	2.0·10 <sup>-8</sup>	3.75·10 <sup>-7</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	1.0·10 <sup>-8</sup>			<sup>a</sup> 4.7·10 <sup>-9</sup>			<sup>a</sup> 4.7·10 <sup>-9</sup>		1.0·10 <sup>-8</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.6·10 <sup>-6</sup>	1.0·10 <sup>-5</sup>	7.3·10 <sup>-5</sup>	8.4·10 <sup>-6</sup>	1.2·10 <sup>-6</sup>	2.9·10 <sup>-6</sup>	1.9·10 <sup>-4</sup>	6.4·10 <sup>-6</sup>	7.8·10 <sup>-6</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>b</sup>	EP <sup>b</sup>	5.0·10 <sup>-9</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	1.5·10 <sup>-9</sup>	EP <sup>b</sup>	EP <sup>b</sup>	
<b>Physico-chemical properties</b>											8-12
molecular weight	g.mol <sup>-1</sup>	390.54	278.35	222.26	334.5	390.6	446.7	194.2	390.6	148.11	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	2018366	38459	266	3427678	100000000	100000000	73	11481536	40	
melting point	°C	-47	-35	-40.5	-27.4	-46	-46	5	-25	130.8	
vapor pressure (25 °C)	Pa	1.3·10 <sup>-5</sup>	2.7·10 <sup>-3</sup>	8.2·10 <sup>-2</sup>	6.7·10 <sup>-4</sup>	1.3·10 <sup>-4</sup>	6.7·10 <sup>-5</sup>	2.5·10 <sup>-1</sup>	1.3·10 <sup>-5</sup>	3.8·10 <sup>-2</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	0.003	11.2	1090	0.05	0.001	0.001	4100	0.0005	6200	
<b>Partition coefficients</b>											12-15
Henry's law constant (25 °C)	Pa.m <sup>-3</sup> .mol <sup>-1</sup>		4.6·10 <sup>-2</sup>	4.8·10 <sup>-2</sup>				1.1·10 <sup>-2</sup>		6.2·10 <sup>-4</sup>	
organic carbon partition coefficient	1.kg <sup>-1</sup>	275423	1380	537	52481		213796	214		36	
<b>Degradation rates</b>											16-19
reaction half-life in air	d	0.7	1.9	5.3	1.6	1.6	1.6	28.0	1.1	121	
hydrolysis in water, soil, sediment (pH=6; 12 °C)	d								2·10 <sup>4</sup>		5·10 <sup>-3</sup>
hydrolysis in water, soil, sediment (pH=7; 12 °C)	d								2·10 <sup>3</sup>		3·10 <sup>-2</sup>
hydrolysis in water, soil, sediment (pH=8; 12 °C)	d								23		5·10 <sup>-2</sup>
biodegradation in surface water (12 °C)	d	19	7	23	35	27	27	5	24	5	
biodegradation in soil (12 °C)	d	19	12	23	60	47	47	5	24	5	
aerobic biodegradation in sediment (12 °C)	d	19	8	23	60	47	47	5	24	5	
anaerobic biodegradation in sediment (12 °C)	d	223	12	138	1092	853	853	18	447	18	
metabolism in plant tissue	d	4									
<b>Exposure assessment</b>											8, 9, 20, 21
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>	300	167	117		92	10	11			
plant conductance	m.s <sup>-1</sup>		4.03·10 <sup>-4</sup>								
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>			0.002							
bioavailability for oral uptake	-	0.5	0.85	0.5	0.5	0.5	0.5	0.5	0.5	0.5	

<sup>a</sup> Default values for HLV<sub>oral</sub> and HLV<sub>inh</sub>, as proposed by RIVM et al. (1994); <sup>b</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> Janus et al. (1994); <sup>2</sup> USEPA (1998b); <sup>3</sup> Environmental Defense fund (1999); <sup>4</sup> Guinée et al. (1996a); <sup>5</sup> Vermeire et al. (1991); <sup>6</sup> RIZA (1999); <sup>7</sup> Huijbregts (1999); <sup>8</sup> Mackay et al. (1995); <sup>9</sup> Staples et al. (1997); <sup>10</sup> USEPA (1998a); <sup>11</sup> Howard (1989); <sup>12</sup> Sabljic et al. (1995); <sup>13</sup> Bockting et al. (1993); <sup>14</sup> Lide (1993); <sup>15</sup> Slooff (1994); <sup>16</sup> Howard et al. (1991); <sup>17</sup> RIVM et al. (1998); <sup>18</sup> Komossa et al. (1995); <sup>19</sup> Peijnenburg et al. (1991); <sup>20</sup> Owen (1990); <sup>21</sup> Riederer (1995)

Table B.5: Substance-specific input parameters for polycyclic aromatics

Compound name	Unit	Naphthalene	Anthracene	Phenanthrene	Fluoranthene	Benzo[a]-anthracene	Chrysene	Benzo[k]-fluoranthene	Benzo[a]pyrene	Benzo[ghi]perylene	Indeno[1,2,3-cd]pyrene	Carcinogenic PAH (total)	Source
CAS nr.	-	91-20-3	120-12-7	85-01-8	206-44-0	56-55-3	218-01-9	207-08-9	50-32-8	191-24-2	193-39-5		
<b>Effect assessment</b>													1-5
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	5.0·10 <sup>-8</sup>	3.0·10 <sup>-7</sup>										6.3·10 <sup>-11</sup>
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	3.0·10 <sup>-9</sup>											1.2·10 <sup>-14</sup>
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	4.2·10 <sup>-7</sup>	3.34·10 <sup>-8</sup>	3.2·10 <sup>-6</sup>	2.4·10 <sup>-7</sup>	1.0·10 <sup>-8</sup>	3.4·10 <sup>-7</sup>	3.6·10 <sup>-9</sup>	5.0·10 <sup>-9</sup>	3.0·10 <sup>-8</sup>	1.8·10 <sup>-8</sup>		1.15·10 <sup>-7</sup>
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	7.0·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	8.0·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>		9.5·10 <sup>-8</sup>
<b>Physico-chemical properties</b>													2, 6-10
molecular weight	g.mol <sup>-1</sup>	128.19	178.2	178.2	202.3	228.3	228.3	252.32	252.3	268.36	276.34		239
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	2344	34674	37154	165959	812830	316228	1000000	1817730	3162278	3837072		1142279
melting point	°C	80.5	216.2	101	111	160	255	217	175	277	162.5		195.6
vapor pressure (25 °C)	Pa	10.4	1.0·10 <sup>-3</sup>	2.0·10 <sup>-2</sup>	1.2·10 <sup>-3</sup>	2.8·10 <sup>-5</sup>	5.7·10 <sup>-7</sup>	5.2·10 <sup>-8</sup>	7.1·10 <sup>-7</sup>	1.4·10 <sup>-8</sup>	1.3·10 <sup>-8</sup>		2.5·10 <sup>-4</sup>
solubility (25 °C)	mg.l <sup>-1</sup>	31	0.045	1.1	0.26	1.1·10 <sup>-2</sup>	2.0·10 <sup>-3</sup>	8.0·10 <sup>-4</sup>	2.1·10 <sup>-3</sup>	2.6·10 <sup>-4</sup>	6.2·10 <sup>-2</sup>		5.9·10 <sup>-2</sup>
<b>Partition coefficients</b>													7, 11-13
Henry's law constant (25 °C)	Pa.m <sup>-3</sup> .mol <sup>-1</sup>	49							9.2·10 <sup>-2</sup>				6.8·10 <sup>-3</sup>
organic carbon partition coefficient	l.kg <sup>-1</sup>	933	23988	22387	41687	199526							107000
<b>Degradation rates</b>													7, 14-18
reaction half-life in air	d		1.23	1.26	0.5	0.07	0.2	0.3	0.035	008	0.16		0.25
hydroxyl radical reaction in air (-10 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	3.2·10 <sup>-11</sup>											
hydroxyl radical reaction in air (12 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	2.5·10 <sup>-11</sup>											
hydroxyl radical reaction in air (25 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	2.7·10 <sup>-11</sup>											
biodegradation in surface water (12 °C)	d	6	264	99	433	459	1062	2434	303	1080	1154		1271
biodegradation in soil (12 °C)	d	49	264	99	433	459	1062	2434	387	1080	1154		1276
aerobic biodegradation in sediment (12 °C)	d	6	264	99	433	459	1062	2434	303	1080	1154		1271
anaerobic biodegradation in sediment (12 °C)	d	140	1058	394	1731	1837	4249	9735	1213	4320	4617		5082
metabolism in plant tissue	d								1.8				
<b>Exposure assessment</b>													6, 7, 19-23
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	398	912	2630		10000			312				18565
bioconcentration factor from porewater to roots	l.kg (wwt) <sup>-1</sup>	30		2700	65								
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>								0.015				
bioavailability for inhalation	-								0.29				0.29
bioavailability for oral uptake	-	1							0.5				0.5

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> Vermeire (1993); <sup>2</sup> WHO-EURO (1996); <sup>3</sup> USEPA (1998b); <sup>4</sup> RIZA (1999); <sup>5</sup> Huijbregts (1999); <sup>6</sup> Mackay et al. (1992b); <sup>7</sup> McKone et al. (1995); <sup>8</sup> Cambridgesoft Corporation (1998); <sup>9</sup> Spectrum Laboratories (1998); <sup>10</sup> Environmental Science Center, 1998; <sup>11</sup> Lide (1993); <sup>12</sup> Sabljic et al. (1995); <sup>13</sup> Bockting et al. (1993); <sup>14</sup> Howard et

al. (1991);<sup>15</sup> RIVM et al. (1998);<sup>16</sup> Atkinson (1986);<sup>17</sup> Kwok et al. (1994);<sup>18</sup> Komossa et al. (1995);<sup>19</sup> Nendza (1991);<sup>20</sup> Devillers et al. (1996);<sup>21</sup> Polder et al. (1995);<sup>22</sup> Dowdy & McKone (1997);<sup>23</sup> Owen (1990)

Table B.6: Substance-specific input parameters for halogenated non-aromatics

Compound name	Unit	dichloro-methane	trichloro-methane	tetrachloro-methane	1,2-dichloro-ethane	1,1,1-trichloro-ethane	trichloroethylene	tetrachloroethylene	vinyl-chloride	hexachlorobutadiene	Source
CAS nr.	-	75-09-2	67-66-3	56-23-5	107-06-2	71-55-6	79-01-6	127-18-4	75-01-4	87-68-3	
<b>Effects assessment</b>											1-8
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>1</sup>	6.0·10 <sup>-8</sup>	1.5·10 <sup>-8</sup>	4.0·10 <sup>-9</sup>	1.4·10 <sup>-10</sup>		5.4·10 <sup>-7</sup>	1.6·10 <sup>-8</sup>	3.51·10 <sup>-11</sup>	2.0·10 <sup>-10</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	4.5·10 <sup>-7</sup>	1.0·10 <sup>-7</sup>	6.0·10 <sup>-8</sup>	7.0·10 <sup>-7</sup>	3.8·10 <sup>-7</sup>	5.39·10 <sup>-9</sup>	2.5·10 <sup>-7</sup>	1.0·10 <sup>-9</sup>	5.0·10 <sup>-11</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.0·10 <sup>-2</sup>	5.9·10 <sup>-3</sup>	1.1·10 <sup>-3</sup>	1.4·10 <sup>-2</sup>	2.1·10 <sup>-3</sup>	2.4·10 <sup>-3</sup>	3.3·10 <sup>-4</sup>	8.2·10 <sup>-3</sup>	5.0·10 <sup>-9</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	4.6·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>											9, 10
molecular weight	g.mol <sup>-1</sup>	84.94	119.38	153.82	98.96	133.41	131.39	165.83	62.5	260.76	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	17.8	93.3	436.5	28	309	309	377	24	50119	
melting point	°C	-95	-63.5	-22.9	-35.36	-30.41	-83.5	-19	-153.8	-21	
vapor pressure (25 °C)	Pa	26222	26244	15250	10740	16500	9664	2560	354600	20	
solubility (25 °C)	mg.l <sup>-1</sup>	13200	8200	800	8514	1495	1437	257	2763	3.2	
<b>Partition coefficients</b>											9-13
Henry's law constant (25 °C)	Pa.m <sup>3</sup> .mol <sup>-1</sup>	285.7	413.3	3248.3	117	2015	1028	1742	2192.8		
organic carbon partition coefficient	l.kg <sup>-1</sup>	36	65	34	42	66.1	110	240			
<b>Degradation rates</b>											9, 10, 14-17
reaction half-life in air	d			4000	73.2				2.4	716	
hydroxyl radical reaction in air (-10 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	8.8·10 <sup>-14</sup>	6.4·10 <sup>-14</sup>			5.6·10 <sup>-15</sup>	2.8·10 <sup>-12</sup>	9.7·10 <sup>-14</sup>			
hydroxyl radical reaction in air (12 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	1.2·10 <sup>-13</sup>	8.7·10 <sup>-14</sup>			9.2·10 <sup>-15</sup>	2.5·10 <sup>-12</sup>	1.4·10 <sup>-13</sup>			
hydroxyl radical reaction in air (25 °C)	cm <sup>3</sup> .molec <sup>-1</sup> .sec <sup>-1</sup>	1.4·10 <sup>-13</sup>	1.0·10 <sup>-13</sup>			1.2·10 <sup>-14</sup>	2.4·10 <sup>-12</sup>	1.7·10 <sup>-13</sup>			
hydrolysis in water, soil, sediment (pH=6, 7, 8; 12 °C)	d			14600	4379	382	560				
biodegradation in surface water (12 °C)	d	24	124	429	435	341	104	122	124	124	
biodegradation in soil (12 °C)	d	24	124	429	102	341	601	444	124	124	
aerobic biodegradation in the sediment zone (12 °C)	d	24	124	429	234	341	313	569	124	124	
anaerobic biodegradation in the sediment zone (12 °C)	d	98	24	24	936	1364	702	702	495	495	
<b>Exposure assessment</b>											9, 10, 18-23
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>		4	30	2	9	26	44	1	9886	
bioavailability for inhalation	-		0.7	0.6			0.6		0.64		
bioavailability for oral uptake	-		0.5	0.9							

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> Vermeire et al. (1991); <sup>2</sup> WHO-EURO (1987); <sup>3</sup> WHO-EURO (1996); <sup>4</sup> USEPA (1998b); <sup>5</sup> Janssen et al. (1995); <sup>6</sup> Janus et al. (1994); <sup>7</sup> RIZA (1999); <sup>8</sup> Huijbregts (1999); <sup>9</sup> Mackay et al. (1993); <sup>10</sup> McKone et al. (1995); <sup>11</sup> Bockting et al. (1993); <sup>12</sup> Van de Plassche & Bockting (1993); <sup>13</sup> Lide (1993); <sup>14</sup> Howard et al. (1991); <sup>15</sup> RIVM et al. (1998); <sup>16</sup> Jeffers et al. (1996); <sup>17</sup> Atkinson (1986); <sup>18</sup> Devillers et al. (1996); <sup>19</sup> Nendza (1991); <sup>20</sup> WHO (1994c); <sup>21</sup> RIVM & TNO (1986); <sup>22</sup> WHO (1985a); <sup>23</sup> Owen (1990)

Table B.7a: Substance-specific input parameters for halogenated aromatics

Compound name	Unit	chloro-benzene	1,2-dichloro-benzene	1,3-dichloro-benzene	1,4-dichloro-benzene	1,2,3-trichloro-benzene	1,2,4-trichloro-benzene	1,3,5-trichloro-benzene	1,2,3,4-tetrachloro-benzene	Source
CAS nr.	-	108-90-7	95-50-1	541-73-1	106-46-7	87-61-6	120-82-1	108-70-3	634-66-2	
<b>Effects assessment</b>										1-5
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.0·10 <sup>-7</sup>	6.0·10 <sup>-7</sup>	<sup>a</sup> 1.0·10 <sup>-9</sup>	2.0·10 <sup>-7</sup>	2.0·10 <sup>-8</sup>	2.0·10 <sup>-8</sup>	2.0·10 <sup>-8</sup>	4.0·10 <sup>-9</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	4.2·10 <sup>-8</sup>	6.0·10 <sup>-8</sup>	<sup>a</sup> 4.7·10 <sup>-9</sup>	6.7·10 <sup>-7</sup>	4.0·10 <sup>-9</sup>	4.0·10 <sup>-9</sup>	4.0·10 <sup>-9</sup>		
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	6.9·10 <sup>-4</sup>	2.7·10 <sup>-4</sup>	2.1·10 <sup>-4</sup>	2.6·10 <sup>-4</sup>	6.4·10 <sup>-5</sup>	7.9·10 <sup>-5</sup>	5.7·10 <sup>-5</sup>	2.3·10 <sup>-5</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	1.8·10 <sup>-7</sup>	EP <sup>b</sup>	EP <sup>b</sup>	9.4·10 <sup>-8</sup>	4.3·10 <sup>-9</sup>	3.8·10 <sup>-8</sup>	1.5·10 <sup>-7</sup>	6.9·10 <sup>-8</sup>	
<b>Physico-chemical properties</b>										6, 7
molecular weight	g.mol <sup>-1</sup>	112.6	147.01	147.01	147.01	181.45	181.45	181.45	215.9	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	631	2512	2512	2925	12589	12589	12589	31623	
melting point	°C	-45.6	-17	-24.9	53.1	53	16.95	64	47.5	
vapor pressure (25 °C)	Pa	1580	196	307	149	28	61	32	5.2	
solubility (25 °C)	mg.l <sup>-1</sup>	484	118	120	72	21	40	5.3	7.8	
<b>Partition coefficients</b>										6-11
Henry's law constant (25 °C)	Pa.m <sup>-3</sup> .mol <sup>-1</sup>	454	219	305	297	306	207	175	70.5	
organic carbon partition coefficient	l.kg <sup>-1</sup>	219	316	302	519	1950	1778	933	9120	
<b>Degradation rates</b>										7, 12, 13
reaction half-life in air	d	18.2	38.2	22.3	50.2	32.1	32.1	32.1	190.8	
biodegradation in surface water (12 °C)	d	176	124	124	124	124	124	124	124	
biodegradation in soil (12 °C)	d	176	782	782	782	124	124	124	124	
aerobic biodegradation in sediment (12 °C)	d	176	124	124	124	124	124	124	124	
anaerobic biodegradation in sediment (12 °C)	d	704	513	513	495	495	495	495	513	
<b>Exposure assessment</b>										6, 7, 14-15
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	74	186	154	196	1778	2239	2042	4074	
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>				75.5		1050			
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>				0.34	0.07	0.07	0.07	0.12	

<sup>a</sup> Default values for HLV<sub>oral</sub> and HLV<sub>inh</sub>, as proposed by RIVM et al. (1994); <sup>b</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> Vermeire et al. (1991); <sup>2</sup> Rademaker et al. (1993); <sup>3</sup> Janus et al. (1994); <sup>4</sup> RIZA (1999); <sup>5</sup> Huijbregts (1999); <sup>6</sup> Mackay et al. (1992a); <sup>7</sup> McKone et al. (1995); <sup>8</sup> Van de Plassche & Bockting (1993); <sup>9</sup> Sabljic et al. (1995); <sup>10</sup> Bockting et al. (1993); <sup>11</sup> Lide (1993); <sup>12</sup> Howard et al. (1991); <sup>13</sup> RIVM et al. (1998); <sup>14</sup> Polder et al. (1998); <sup>15</sup> Dowdy & McKone (1997)

Table B.7b: Substance-specific input parameters for halogenated aromatics

Compound name	Unit	1,2,3,5-tetra-chlorobenzene	1,2,4,5-tetra-chlorobenzene	pentachloro-benzene	hexachloro-benzene	2-chloro-phenol	2,4-dichloro-phenol	2,4,5-tri-chlorophenol	2,4,6-tri-chlorophenol	Source
CAS nr.	-	634-90-2	95-94-3	608-93-5	118-74-1	95-57-8	120-83-2	95-95-4	88-06-2	
<b>Effects assessment</b>										1-6
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	4.0·10 <sup>-9</sup>	4.0·10 <sup>-9</sup>	8.0·10 <sup>-10</sup>	5.7·10 <sup>-13</sup>	5.0·10 <sup>-9</sup>	3.0·10 <sup>-9</sup>	1.0·10 <sup>-7</sup>	8.6·10 <sup>-11</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>				2.0·10 <sup>-12</sup>	1.8·10 <sup>-8</sup>			3.0·10 <sup>-10</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.2·10 <sup>-5</sup>	2.6·10 <sup>-5</sup>	7.5·10 <sup>-6</sup>	2.4·10 <sup>-6</sup>	3.0·10 <sup>-6</sup>	5.8·10 <sup>-6</sup>	4.8·10 <sup>-6</sup>	1.3·10 <sup>-5</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	3.5·10 <sup>-9</sup>	3.0·10 <sup>-9</sup>	8.3·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	2.3·10 <sup>-8</sup>	2.6·10 <sup>-8</sup>	
<b>Physico-chemical properties</b>										7-9
molecular weight	g.mol <sup>-1</sup>	215.9	215.9	250.3	284.79	128.56	163	197.45	197.45	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	31623	31623	100000	316228	148	1589	5273	4921	
melting point	°C	54.5	140	86	228	9	44	69	69.5	
vapor pressure (25 °C)	Pa	9.8	0.7	2.2·10 <sup>-1</sup>	2.3·10 <sup>-3</sup>	132	12	2.5	1.25	
solubility (25 °C)	mg.l <sup>-1</sup>	3.6	1.3	0.65	5.0·10 <sup>-3</sup>	24650 <sup>b</sup>	4500 <sup>b</sup>	948 <sup>b</sup>	434 <sup>b</sup>	
dissociation constant	-					8.49	7.68	7.43	7.42	
<b>Partition coefficients</b>										10-13
Henry's law constant (25 °C)	Pa.m <sup>-3</sup> .mol <sup>-1</sup>	118	94	7.05·10 <sup>1</sup>	78.2	5.7·10 <sup>2</sup>			6.2·10 <sup>-3</sup>	
organic carbon partition coefficient	l.kg <sup>-1</sup>	2399	8128	4169	10965	136	295	1413	1047	
<b>Degradation rates</b>										7, 14-16
reaction half-life in air	d	190.8	190.8	271.9	938.3	2.0	5.3	7.6	30.8	
biodegradation in surface water (12 °C)	d	124	124	451	2481	53	8	220	39	
biodegradation in soil (12 °C)	d	124	124	451	2481	6	39	220	39	
aerobic biodegradation in sediment (12 °C)	d	124	124	451	2481	53	8	220	39	
anaerobic biodegradation in sediment (12 °C)	d	513	513	1805	9925	212	42	836	967	
metabolism in plant tissue	d				10.2					
<b>Exposure assessment</b>										7, 8, 17-22
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	2042	6095	5754	13490	214		1905	676	
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>		2.4·10 <sup>3</sup>	1.5·10 <sup>4</sup>	2.0·10 <sup>4</sup>					
plant conductance	m.s <sup>-1</sup>				6.76·10 <sup>-4</sup>					
bioconcentration factor from porewater to roots	l.kg (wwt) <sup>-1</sup>						130			
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>	0.12	0.12	0.05	0.03		0.09			
biotransfer factor for meat	d.kg <sup>-1</sup>				0.05					
biotransfer factor for milk	d.kg <sup>-1</sup>				0.009					

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>1</sup> Vermeire et al. (1991); <sup>2</sup> Environmental Defense Fund (1999); <sup>3</sup> USEPA (1998); <sup>4</sup> Janus et al. (1994); <sup>5</sup> RIZA (1999); <sup>6</sup> Huijbregts (1999); <sup>7</sup> Mackay et al. (1992a); <sup>8</sup> Mackay et al. (1997); <sup>9</sup> Mackay et al. (1995); <sup>10</sup> Van de Plassche & Bockting (1993); <sup>11</sup> Sabljic et al. (1995); <sup>12</sup> Bockting et al. (1993); <sup>13</sup> Lide (1993); <sup>14</sup> Howard et al. (1991); <sup>15</sup> RIVM et al. (1998); <sup>16</sup> Komoša et al. (1995); <sup>17</sup> Nendza (1991); <sup>18</sup> Polder et al. (1995); <sup>19</sup> Polder et al. (1998); <sup>20</sup> Dowdy & McKone (1997); <sup>21</sup> Travis & Arms (1988); <sup>22</sup> Riederer (1995)

Table B.7c: Substance-specific input parameters for halogenated aromatics

Compound name	Unit	2,3,4,6-tetra-chlorophenol	pentachloro-phenol	Benzyl-chloride	3-chloro-aniline	4-chloro-aniline	3,4-chloro-aniline	1-chloro-4-nitrobenzen-e	pentachloro-nitrobenzene	2,3,7,8-TCDD	Source
CAS nr.	-	58-90-2	87-86-5	100-44-7	108-42-9	106-47-8	95-76-1	100-00-5	82-68-8	1746-01-6	
<b>Effects assessment</b>											1-10
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.0·10 <sup>-8</sup>	3.0·10 <sup>-8</sup>	5.7·10 <sup>-12</sup>	9.0·10 <sup>-12</sup>	9.0·10 <sup>-12</sup>	2.0·10 <sup>-9</sup>	5.56·10 <sup>-11</sup>	7.0·10 <sup>-9</sup>	1.0·10 <sup>-15</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>		1.0·10 <sup>-7</sup>		4.0·10 <sup>-11</sup>	4.0·10 <sup>-11</sup>					
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.4·10 <sup>-6</sup>	3.5·10 <sup>-6</sup>	1.3·10 <sup>-6</sup>	1.3·10 <sup>-6</sup>	8.0·10 <sup>-7</sup>	8.0·10 <sup>-7</sup>	3.2·10 <sup>-6</sup>	2.9·10 <sup>-7</sup>	1.2·10 <sup>-12</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	1.1·10 <sup>-8</sup>	EP <sup>a</sup>	3.1·10 <sup>-8</sup>	EP <sup>a</sup>	1.5·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>											10-15
molecular weight	g.mol <sup>-1</sup>	231.89	266.34	126.58	127.57	127.57	162.06	157.56	295.3	322	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	42975	282026	200	76	68	468	245	44000	8.13·10 <sup>6</sup>	
melting point	°C	70	190	-45.5	-10	70	71.5	83	146	305	
vapor pressure (25 °C)	Pa	2.8·10 <sup>-1</sup>	4.15·10 <sup>-3</sup>	163.7	9.5	2.3	1.3	2.0	6.6·10 <sup>-3</sup>	2.0·10 <sup>-7</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	183 <sup>b</sup>	14 <sup>b</sup>	528	5440 <sup>b</sup>	3000 <sup>b</sup>	92 <sup>b</sup>	342	0.44	2.5·10 <sup>-5</sup>	
dissociation constant	-	5.38	4.92		3.5	4.0	2.5				
<b>Partition coefficients</b>											9, 15-18
Henry's law constant (25 °C)	Pa.m <sup>3</sup> .mol <sup>-1</sup>						1.1		3.6		
organic carbon partition coefficient	l.kg <sup>-1</sup>	2239	32359		355	180	977		23000	1.8·10 <sup>6</sup>	
<b>Degradation rates</b>											11, 14, 19-25
reaction half-life in air	d	25	34.8	5.5	0.2	0.2	0.4	272	2198		
hydroxyl radical reaction in air (-10 °C)	cm <sup>3</sup> ..molec <sup>-1</sup> .sec <sup>-1</sup>									4.0·10 <sup>-13</sup>	
hydroxyl radical reaction in air (12 °C)	cm <sup>3</sup> ..molec <sup>-1</sup> .sec <sup>-1</sup>									6.6·10 <sup>-13</sup>	
hydroxyl radical reaction in air (25 °C)	cm <sup>3</sup> ..molec <sup>-1</sup> .sec <sup>-1</sup>									8.6·10 <sup>-13</sup>	
hydrolysis in water, soil, sediment (PH=6, 7, 8; 12 °C)	d			5							
biodegradation in surface water (12 °C)	d	120	112	24	30	30	1000	451	673	622	
biodegradation in soil (12 °C)	d	120	112	24	100	100	1000	451	673	3194	
aerobic biodegradation in sediment (12 °C)	d	120	112	24	100	100	1000	451	673	622	
anaerobic biodegradation in sediment (12 °C)	d	478	441	98	400	400	4000	1805	31	1757	
metabolism in plant tissue	d					0.6	0.7				
photodegradation upon plant tissue	d									6.2	
<b>Exposure assessment</b>											14, 15, 26-32
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>	447	268		4	5	30		201	12314	
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>									3.2·10 <sup>7</sup>	
plant conductance	m.s <sup>-1</sup>		1.91·10 <sup>-3</sup>								
transpiration stream concentration factor	-	0.04 <sup>c</sup>	0.04 <sup>c</sup>		0.04 <sup>c</sup>	0.04 <sup>c</sup>	0.04 <sup>c</sup>				
bioconcentration factor from porewater to roots	1.kg(wwt) <sup>-1</sup>	0.7 <sup>c</sup>	1.2		0.7 <sup>c</sup>	0.7 <sup>c</sup>	0.7 <sup>c</sup>			780	
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>						0.35			3·10 <sup>-2</sup>	
biotransfer factor for meat										1.8·10 <sup>-3</sup>	0.11
biotransfer factor for milk	d.kg <sup>-1</sup>									1.6·10 <sup>-4</sup>	8·10 <sup>-3</sup>
bioavailability for oral uptake	-										0.5

<sup>a</sup>EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>c</sup> assumed default value for dissociating acids at environmental pH of 7, based on Briggs et al. (1987); <sup>1</sup> USEPA (1998b); <sup>2</sup> Vermeire et al. (1991); <sup>3</sup> Environmental Defense Fund (1999); <sup>4</sup> Janssen et al. (1998); <sup>5</sup> Janus et al. (1994); <sup>6</sup> Health Council of the Netherlands (1996); <sup>7</sup> Lu (1995); <sup>8</sup> RIZA (1999); <sup>9</sup> Huijbregts (1999); <sup>10</sup> Guinée et al. (1996a); <sup>11</sup> Howard (1989); <sup>12</sup> Verschueren (1996); <sup>13</sup> Mackay et al. (1992b); <sup>14</sup> Mackay et al. (1995); <sup>15</sup> Mackay et al. (1997); <sup>16</sup> Sabljic et al. (1995); <sup>17</sup> Bockting et al. (1993); <sup>18</sup> Lide (1993); <sup>19</sup> Howard et al. (1991); <sup>20</sup> RIVM et al. (1998); <sup>21</sup> Jager et al. (1997); <sup>22</sup> Syracuse Research Corporation (1993); <sup>23</sup> Brubaker & Hites (1997); <sup>24</sup> Trapp & Matthies (1995); <sup>25</sup> Komoša et al. (1995); <sup>26</sup> Liem et al. (1993); <sup>27</sup> Polder et al. (1995); <sup>28</sup> Polder et al. (1998); <sup>29</sup> Dowdy et al. (1996); <sup>30</sup> Dowdy & McKone (1997); <sup>31</sup> WHO (1989a); <sup>32</sup> Riederer (1995)

Table B.8a: Substance-specific input parameters for pesticides

Compound name	Unit	Acephate	Aldicarb	Aldrin	Anilazine	Atrazine	Azinphosethyl	Azinphos-methyl	Benomyl	Bentazone	Bifenthrin	Source
CAS nr.	-	30560-19-1	116-06-3	309-00-2	101-05-3	1912-24-9	2642-71-9	86-50-0	17804-35-2	25057-89-0	82657-04-3	
<b>Effects assessment</b>												1-7
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>1</sup>	3.0·10 <sup>-8</sup>	3.0·10 <sup>-9</sup>	1.0·10 <sup>-10</sup>	1.0·10 <sup>-7</sup>	3.5·10 <sup>-8</sup>	2.5·10 <sup>-10</sup>	5.0·10 <sup>-9</sup>	1.0·10 <sup>-7</sup>	1.0·10 <sup>-7</sup>	2.0·10 <sup>-8</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>							2.0·10 <sup>-10</sup>				
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	6.4·10 <sup>-6</sup>	2.0·10 <sup>-8</sup>	2.9·10 <sup>-8</sup>	2.0·10 <sup>-7</sup>	2.9·10 <sup>-6</sup>	1.1·10 <sup>-8</sup>	1.2·10 <sup>-8</sup>	1.5·10 <sup>-7</sup>	6.4·10 <sup>-5</sup>	1.1·10 <sup>-9</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	EP <sup>a</sup>	2.0·10 <sup>-9</sup>	EP <sup>a</sup>	7.0·10 <sup>-9</sup>	EP <sup>a</sup>	2.0·10 <sup>-8</sup>	1.5·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>												2, 8, 9
molecular weight	g.mol <sup>-1</sup>	183.2	190.25	364.93	275.5	215.68	345.4	317.34	290.3	240.3	422.9	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	0.1	13	1000	6300	562	1514	500	200	631	1000000	
melting point	°C	85.4	99.5	104	159.5	174	50	73.5	140	138	58	
vapor pressure (25 °C)	Pa	2.3·10 <sup>-4</sup>	4.0·10 <sup>-3</sup>	5.0·10 <sup>-3</sup>	8.2·10 <sup>-7</sup>	4.0·10 <sup>-5</sup>	4.5·10 <sup>-4</sup>	3.0·10 <sup>-5</sup>	1.3·10 <sup>-8</sup>	6.5·10 <sup>-4</sup>	2.4·10 <sup>-5</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	818000	6000	2·10 <sup>-2</sup>	8	30	4.8	30	2	611 <sup>b</sup>	0.1	
dissociation constant	-									5		
<b>Partition coefficients</b>												8, 9
organic carbon partition coefficient	l.kg <sup>-1</sup>	2	17	410	1000	155		1300	1900	33	223872	
<b>Degradation rates</b>												2, 9-15
reaction half-life in air	d		2.4·10 <sup>-1</sup>	2.3·10 <sup>-1</sup>				5.4·10 <sup>-2</sup>	6.7·10 <sup>-2</sup>			
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	5.1·10 <sup>-11</sup>			1.1·10 <sup>-10</sup>	1.5·10 <sup>-10</sup>	1.7·10 <sup>-10</sup>			1.1·10 <sup>-10</sup>	3.4·10 <sup>-11</sup>	
hydrolysis in surface water, soil, sediment (PH=6; 12 °C)	d		975	1325	84		21	98	12			
hydrolysis in surface water, soil, sediment (PH=7; 12 °C)	d		1083	1325	57		471	87	12			
hydrolysis in surface water, soil, sediment (PH=8; 12 °C)	d		565	1325	14		167	42	12			
biodegradation in surface water (12 °C)	d	91	148	194	2	780	30	5	30	30	4	
biodegradation in soil (12 °C)	d	4	148	194	2	71	49	79	117	84	382	
aerobic biodegradation in sediment (12 °C)	d	4	148	194	2	71	49	79	117	84	382	
anaerobic biodegradation in sediment (12 °C)	d	15	346	5	7	284	195	314	467	335	1528	
metabolism in plant tissue	d		0.6									
<b>Exposure assessment</b>												9, 16-22
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	1	42	5956		8						
plant conductance	m.s <sup>-1</sup>					1.18·10 <sup>-3</sup>						
transpiration stream concentration factor	-									0.04 <sup>c</sup>		
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>		9.4·10 <sup>-1</sup>			1.9				0.7 <sup>c</sup>		
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>		3.7·10 <sup>-1</sup>	2.5·10 <sup>-2</sup>		2.6·10 <sup>-1</sup>			3.6·10 <sup>-1</sup>			
biotransfer factor for meat	d.kg <sup>-1</sup>		9.4·10 <sup>-5</sup>	8.4·10 <sup>-2</sup>				7.9·10 <sup>-4</sup>				
biotransfer factor for milk	d.kg <sup>-1</sup>			1.8·10 <sup>-2</sup>								
bioavailability for inhalation	-			0.3								
bioavailability for oral uptake	-			1								

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>c</sup> assumed default value for dissociating acids at environmental pH of 7, based on Briggs et al. (1987); <sup>1</sup> WHO/FAO (1998); <sup>2</sup> Tomlin (1994); <sup>3</sup> Vermeire et al. (1991); <sup>4</sup> USEPA (1998b); <sup>5</sup> RIZA (1999); <sup>6</sup> Huijbregts (1999); <sup>7</sup> Janssen et al. (1995); <sup>8</sup> Crommentuijn et al. (1997b); <sup>9</sup> Mackay et al. (1997); <sup>10</sup> Howard et al. (1991); <sup>11</sup> RIVM et al. (1998); <sup>12</sup> Jager et al. (1997); <sup>13</sup> Syracuse Research Corporation (1993); <sup>14</sup> Linders et al. (1994); <sup>15</sup> Howard (1991); <sup>16</sup> Polder et al. (1995); <sup>17</sup> Briggs et al. (1982); <sup>18</sup> Dowdy & McKone (1997); <sup>19</sup> Dowdy et al. (1996); <sup>20</sup> Garten & Trabalka (1983); <sup>21</sup> WHO (1989c); <sup>22</sup> Riederer (1995)

Table B.8b: Substance-specific input parameters for pesticides

Compound name	Unit	Captafol	Captan	Carbaryl	Carbendazim	Carbofuran	Chlordane	Chlorfenvinphos	Chloridazone	Chlorothalonil	Source
CAS nr.	-	2425-06-1		63-25-2	10605-21-7	1563-66-2	57-74-9	470-90-6	1698-60-8	1897-45-6	
<b>Effects assessment</b>											1-7
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>1</sup>	2.0·10 <sup>-9</sup>	1.0·10 <sup>-7</sup>	3.0·10 <sup>-9</sup>	3.0·10 <sup>-8</sup>	2.0·10 <sup>-9</sup>	5.0·10 <sup>-10</sup>	5.0·10 <sup>-10</sup>	1.6·10 <sup>-7</sup>	3.0·10 <sup>-8</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>						2.0·10 <sup>-11</sup>				
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.8·10 <sup>-8</sup>	2.2·10 <sup>-8</sup>	2.3·10 <sup>-7</sup>	2.0·10 <sup>-7</sup>	2.0·10 <sup>-7</sup>	1.5·10 <sup>-9</sup>	3.3·10 <sup>-6</sup>	7.3·10 <sup>-5</sup>	8.8·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	1.5·10 <sup>-8</sup>	3.0·10 <sup>-9</sup>	3.7·10 <sup>-8</sup>	1.8·10 <sup>-9</sup>	1.2·10 <sup>-9</sup>	7.0·10 <sup>-9</sup>	2.2·10 <sup>-8</sup>	2.3·10 <sup>-8</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>											2, 8, 9
molecular weight	g.mol <sup>-1</sup>	349.1	300.6	201.22	191.19	221.3	409.8	359.56	221.6	265.89	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	3273	200	229	33	209	1000000	6607	14	437	
melting point	°C	160.5	178	142	304.5	151	104	-19	207	250.5	
vapor pressure (25 °C)	Pa	1.0·10 <sup>-10</sup>	1.1·10 <sup>-5</sup>	2.7·10 <sup>-5</sup>	6.5·10 <sup>-8</sup>	8.0·10 <sup>-5</sup>	5.2·10 <sup>-4</sup>	1.0·10 <sup>-4</sup>	7	1.3·10 <sup>-1</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	1.5	5.1	120.0	8.0 <sup>b</sup>	351.0	0.1	124.0	360.0	0.6	
dissociation constant	-				4.48						
<b>Partition coefficients</b>											8, 9
organic carbon partition coefficient	1.kg <sup>-1</sup>	2090	200	180	410	43	316228	480	81	1585	
<b>Degradation rates</b>											9-16
reaction half-life in air	d		0.08	0.2		0.2	1.3				
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	8.9·10 <sup>-11</sup>			2.1·10 <sup>-10</sup>			5.3·10 <sup>-11</sup>	4.8·10 <sup>-11</sup>	6.2·10 <sup>-15</sup>	
hydrolysis in surface water, soil, sediment (PH=6; 12 °C)	d		1	262		110					
hydrolysis in surface water, soil, sediment (PH=7; 12 °C)	d		0.3	26		110					
hydrolysis in surface water, soil, sediment (PH=8; 12 °C)	d		0.3	3		101					
biodegradation in surface water (12 °C)	d	1000	38	12	105	30	1092	36	262	12	
biodegradation in soil (12 °C)	d	1000	38	12	209	26	1092	63	52	37	
aerobic biodegradation in sediment (12 °C)	d	10000	38	12	209	26	1092	63	52	37	
anaerobic biodegradation in sediment (12 °C)	d	40000	153	49	837	46	5	251	209	148	
metabolism in plant tissue	d			2							
<b>Exposure assessment</b>											9, 17-22
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>		10	17		118	5594				
transpiration stream concentration factor	-				8.8·10 <sup>-1</sup>						
bioconcentration factor from porewater to roots	1.kg(wwt) <sup>-1</sup>					1.3					
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>						1.7·10 <sup>-2</sup>				
biotransfer factor for meat	d.kg <sup>-1</sup>						1.5·10 <sup>-2</sup>				
biotransfer factor for milk	d.kg <sup>-1</sup>						4.6·10 <sup>-4</sup>				
bioavailability for oral uptake	-				8.2·10 <sup>-1</sup>				3.0·10 <sup>-1</sup>		

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>1</sup> FAO/WHO (1998); <sup>2</sup> Tomlin (1994); <sup>3</sup> USEPA (1998b); <sup>4</sup> Janssen et al. (1995); <sup>5</sup> RIZA (1999); <sup>6</sup> Huijbregts (1999); <sup>7</sup> USEPA (1998b); <sup>8</sup> Crommentuijn et al. (1997b); <sup>9</sup> Mackay et al. (1997); <sup>10</sup> Howard et al. (1991); <sup>11</sup> RIVM et al. (1998); <sup>12</sup> Jager et al. (1997); <sup>13</sup> Syracuse Research Corporation (1993); <sup>14</sup> Linders et al. (1994); <sup>15</sup> Howard (1991); <sup>16</sup> Van Rijn et al. (1995); <sup>17</sup> Siebaldi et al. (1997); <sup>18</sup> Polder et al. (1995); <sup>19</sup> Dowdy & McKone (1997); <sup>20</sup> Dowdy et al. (1996); <sup>21</sup> WHO (1993a); <sup>22</sup> WHO (1996b)

Table B.8c: Substance-specific input parameters for pesticides

Compound name	Unit	Chlorpropham	Chlorpyriphos	Coumaphos	Cyanazine	Cypermethrin	Cyromazine	2,4-D	DDT	Deltamethrin	Demeton	Source
CAS nr.	-	101-21-3	2921-88-2	56-72-4	21725-46-2	52315-07-8	66215-27-8	94-75-7	50-29-3	52918-63-5	8065-48-3	
<b>Effects assessment</b>												
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	2.0·10 <sup>-7</sup>	1.0·10 <sup>-8</sup>	2.5·10 <sup>-10</sup>	6.0·10 <sup>-9</sup>	5.0·10 <sup>-8</sup>	2.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	2.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	4.0·10 <sup>-11</sup>	1-6
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	3.8·10 <sup>-5</sup>	7.4·10 <sup>-10</sup>	7.4·10 <sup>-10</sup>	5.0·10 <sup>-8</sup>	1.3·10 <sup>-10</sup>	4.5·10 <sup>-7</sup>	9.9·10 <sup>-6</sup>	5.0·10 <sup>-9</sup>	3.0·10 <sup>-10</sup>	1.4·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	1.0·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	1.2·10 <sup>-8</sup>	2.8·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>												
molecular weight	g.mol <sup>-1</sup>	213.65	350.6	362.8	240.7	416.3	166.2	221.04	354.5	505.2	258.34	3, 7, 8
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	3240	8.3·10 <sup>4</sup>	11350	166	4.0·10 <sup>6</sup>	1	646	1.6·10 <sup>6</sup>	39810	16	
melting point	°C	40.9	41.5	95	166.75	80.5	221	140.5	108.75	99.5	liquid	
vapor pressure (25 °C)	Pa	1.0·10 <sup>-3</sup>	2.3·10 <sup>-3</sup>	1.8·10 <sup>-5</sup>	2.1·10 <sup>-7</sup>	1.9·10 <sup>-7</sup>	4.5·10 <sup>-7</sup>	8.0·10 <sup>-5</sup>	2.0·10 <sup>-5</sup>	1.0·10 <sup>-5</sup>	3.5·10 <sup>-2</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	89	7.3·10 <sup>-1</sup>	1.6	171	4.0·10 <sup>-3</sup>	13000	400 <sup>b</sup>	5.5·10 <sup>-3</sup>	2.0·10 <sup>-3</sup>	60	
dissociation constant	-							3.0				
<b>Partition coefficients</b>												7, 8
organic carbon partition coefficient	l.kg <sup>-1</sup>	347	7590	2089	178	389		46	427000	4.6·10 <sup>5</sup>	71	
<b>Degradation rates</b>												8-15
reaction half-life in air	d		2.6·10 <sup>-1</sup>					4.5·10 <sup>-1</sup>	4.4			
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	4.7·10 <sup>-11</sup>		1.1·10 <sup>-10</sup>	1.3·10 <sup>-10</sup>	3.7·10 <sup>-11</sup>	1.0·10 <sup>-10</sup>			3.9·10 <sup>-11</sup>	1.3·10 <sup>-10</sup>	
hydrolysis in water (PH= 6, 12 °C)	d		209									
hydrolysis in water (PH= 7, 12 °C)	d		92									
hydrolysis in water (PH= 8, 12 °C)	d		3									
biodegradation in water (12 °C)	d	29	37	1000	24	9	269	39	3556	5	30	
hydrolysis in soil (PH=6, 12 °C)	d		2090									
hydrolysis in soil, sediment (PH=7, 12 °C)	d		924									
hydrolysis in sediment (PH=8, 12 °C)	d		26									
biodegradation in soil (12 °C)	d	70	37	1000	24	91	162	39	3556	44	26	
aerobic biodegradation in sediment (12 °C)	d	70	37	10000	49	91	162	39	3556	436	26	
anaerobic biodegradation in sediment (12 °C)	d	279	147	40000	195	362	649	123	70	1744	105	
metabolism in plant tissue	d	1.4						1.8	18.4			
<b>Exposure assessment</b>												8, 15-22
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>		752			813		0.1	16549	457		
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>								6.74·10 <sup>6</sup>			
plant conductance	m.s <sup>-1</sup>							4.88·10 <sup>-4</sup>				
transpiration stream concentration factor	-							0.04				
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>							1	4100			
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>				2.0·10 <sup>-1</sup>				1.2·10 <sup>-2</sup>			
biotransfer factor for meat	d.kg <sup>-1</sup>		7.9·10 <sup>-4</sup>	3.8·10 <sup>-6</sup>				1.0·10 <sup>-5</sup>	3.2·10 <sup>-2</sup>			
biotransfer factor for milk	d.kg <sup>-1</sup>		2.1·10 <sup>-5</sup>	4.7·10 <sup>-6</sup>				3.7·10 <sup>-6</sup>	3.2·10 <sup>-3</sup>			

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>1</sup> Vermeire et al. (1991); <sup>2</sup> FAO/WHO (1998); <sup>3</sup> Environmental Defense Fund (1999); <sup>4</sup> USEPA (1998b); <sup>5</sup> Huijbregts (1999); <sup>6</sup> RIZA (1999); <sup>7</sup> Crommentuijn et al. (1997b); <sup>8</sup> Mackay et al. (1997); <sup>9</sup> Howard et al. (1991); <sup>10</sup> RIVM et al. (1998); <sup>11</sup> Jager et al. (1997); <sup>12</sup> Syracuse Research Corporation

(1993);<sup>13</sup> Linders et al. (1994);<sup>14</sup> Komóćka et al. (1995);<sup>15</sup> Briggs et al. (1987);<sup>16</sup> Tsuda et al. (1997);<sup>17</sup> Polder et al. (1995);<sup>18</sup> Polder et al. (1998);<sup>19</sup> Devillers et al. (1996);<sup>20</sup> Dowdy & McKone (1997);<sup>21</sup> Dowdy et al. (1996);<sup>22</sup> Riederer (1995)

Table B.8d: Substance-specific input parameters for pesticides

Compound name	Unit	Desmetryn	Diazinon	Dichlorprop	Dichlorvos	Dieldrin	Dimethoate	Dinoseb	Dinoterb	Disulfoton	Diuron	Source
CAS nr.	-	1014-69-3	333-41-5	120-36-5	62-73-7	60-57-1	60-51-5	88-85-7	1420-07-1	298-04-4	330-54-1	
<b>Effects assessment</b>												1-6
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	<sup>a</sup> 1.5·10 <sup>-9</sup>	2.0·10 <sup>-9</sup>	5.0·10 <sup>-9</sup>	4.0·10 <sup>-9</sup>	1.0·10 <sup>-10</sup>	2.0·10 <sup>-9</sup>	1.0·10 <sup>-9</sup>	<sup>b</sup> 3.75·10 <sup>-8</sup>	3.0·10 <sup>-10</sup>	2.0·10 <sup>-9</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>		9.0·10 <sup>-9</sup>		5.0·10 <sup>-10</sup>					2.0·10 <sup>-10</sup>		
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.6·10 <sup>-5</sup>	3.7·10 <sup>-8</sup>	4.0·10 <sup>-4</sup>	7.0·10 <sup>-10</sup>	2.9·10 <sup>-8</sup>	2.3·10 <sup>-5</sup>	2.5·10 <sup>-8</sup>	3.4·10 <sup>-8</sup>	2.3·10 <sup>-8</sup>	4.3·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	2.5·10 <sup>-9</sup>	1.9·10 <sup>-9</sup>	EP <sup>c</sup>	EP <sup>c</sup>	3.3·10 <sup>-9</sup>	1.7·10 <sup>-8</sup>	EP <sup>c</sup>	EP <sup>c</sup>	EP <sup>c</sup>	2.4·10 <sup>-9</sup>	
<b>Physico-chemical properties</b>												2, 7, 8
molecular weight	g.mol <sup>-1</sup>	213.3	304.36	235.1	220.98	380.93	229.28	240.2	240.2	274.38	233.1	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	240	2000	2690	28	1.6·10 <sup>5</sup>	6	3630	5495	10500	603	
melting point	°C	85	liquid	116.75	liquid	176.5	52.25	40	126	108	158.5	
vapor pressure (25 °C)	Pa	1.9·10 <sup>-4</sup>	8.0·10 <sup>-3</sup>	4.0·10 <sup>-4</sup>	7.0	5.0·10 <sup>-4</sup>	1.0·10 <sup>-2</sup>	3.2·10 <sup>-1</sup>	2.8·10 <sup>-2</sup>	2.0·10 <sup>-2</sup>	9.2·10 <sup>-5</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	621	60	350 <sup>d</sup>	8000	1.7·10 <sup>-1</sup>	20000	48.5 <sup>d</sup>	4.8 <sup>d</sup>	25	40	
dissociation constant	-			3.25				4.5	4.6			
<b>Partition coefficients</b>												7, 8
organic carbon partition coefficient	1.kg <sup>-1</sup>		437	1000	68	12000	58	195		1320	355	
<b>Degradation rates</b>												2, 7, 9-12
reaction half-life in air	d		0.2				1.0	0.1	3.1		0.1	
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	1.4·10 <sup>-10</sup>		1.1·10 <sup>-11</sup>	9.2·10 <sup>-12</sup>				6.2·10 <sup>-13</sup>		3.4·10 <sup>-11</sup>	
hydrolysis in water, soil, sediment (PH=6, 12 °C)	d		223		5		460				230	
hydrolysis in water, soil sediment (PH=7, 12 °C)	d		316		5		378				230	
hydrolysis in water, soil, sediment (PH=8, 12 °C)	d		317		5		99				230	
biodegradation in water (12 °C)	d	55	47	19	0.8	763	43	127	164	14	47	
biodegradation in soil (12 °C)	d	16	56	21	0.4	763	35	127	17	14	105	
aerobic biodegradation in sediment (12 °C)	d	16	56	21	0.4	763	43	127	17	14	105	
anaerobic biodegradation in sediment (12 °C)	d	63	223	85	2	5	173	14	68	55	420	
metabolism in plant tissue	d						4.2					
<b>Exposure assessment</b>												7, 14-20
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>		83		0.6	6166		32	0.8	208	60	
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>					1.1·10 <sup>6</sup>						
transpiration stream concentration factor	-			0.04 <sup>e</sup>			4.3·10 <sup>-1</sup>	0.04 <sup>e</sup>	0.04 <sup>e</sup>		8.1·10 <sup>-1</sup>	
bioconcentration factor from porewater to roots	1.kg(wwt) <sup>-1</sup>			0.7 <sup>e</sup>		43		0.7 <sup>e</sup>	0.7 <sup>e</sup>		3.1	
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>					1.0·10 <sup>-2</sup>						
biotransfer factor for meat	d.kg <sup>-1</sup>					6.8·10 <sup>-2</sup>						
biotransfer factor for milk	d.kg <sup>-1</sup>					1.2·10 <sup>-2</sup>						

<sup>a</sup> The oral Human Limit Values of Desmetryn is derived by dividing a NOEL for rats (90 days), listed in Tomlin (1994), by a factor 1000 (inter- and intraspecies extrapolation, and extrapolation from subchronic to chronic) (personal assessment); <sup>b</sup> The oral Human Limit Values of Dinoterb is derived by dividing a NOEL for rats (2 years), listed in Tomlin (1994), by a factor 100 (intra- and interspecies extrapolation) (personal assessment); <sup>c</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>d</sup> solubility of the neutral species; <sup>e</sup> assumed default value for dissociating acids at

environmental pH of 7, based on Briggs et al. (1987); <sup>1</sup> FAO/WHO (1998); <sup>2</sup> Tomlin (1994); <sup>3</sup> USEPA (1998b); <sup>4</sup> Environmental Defense Fund (1999); <sup>5</sup> RIZA (1999); <sup>6</sup> Huijbregts (1999); <sup>7</sup> Mackay et al. (1997); <sup>8</sup> Crommentuijn et al. (1997b); <sup>9</sup> Howard et al. (1991); <sup>10</sup> RIVM et al. (1998); <sup>11</sup> Syracuse Research Corporation (1993) <sup>12</sup> Linders et al. (1994); <sup>13</sup> Komôba et al. (1995); <sup>14</sup> Sicbaldi et al. (1997); <sup>15</sup> Nendza (1991); <sup>16</sup> Tsuda et al. (1997); <sup>17</sup> Polder et al. (1995); <sup>18</sup> Polder et al. (1998); <sup>19</sup> Dowdy & McKone (1997); <sup>20</sup> Dowdy et al. (1996)

Table B.8e: Substance-specific input parameters for pesticides

Compound name	Unit	DNOC	Endosulfan	Endrin	Ethoprophos	Fenitrothion	Fentin acetate	Fentin chloride	Fentin hydroxide	Fenthion	Folpet	Source
CAS nr.	-	534-52-1	115-29-7	72-20-8	13194-48-4	122-14-5	900-95-8	639-58-7	76-87-9	55-38-9	133-07-3	
<b>Effects assessment</b>												1-5
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.5·10 <sup>-10</sup>	6.0·10 <sup>-9</sup>	2.0·10 <sup>-10</sup>	4.0·10 <sup>-10</sup>	5.0·10 <sup>-9</sup>	5.0·10 <sup>-10</sup>	5.0·10 <sup>-10</sup>	5.0·10 <sup>-10</sup>	7.0·10 <sup>-9</sup>	1.0·10 <sup>-7</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>	3.5·10 <sup>-10</sup>										
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	2.1·10 <sup>-5</sup>	2.0·10 <sup>-8</sup>	3.0·10 <sup>-9</sup>	6.3·10 <sup>-8</sup>	8.7·10 <sup>-9</sup>	5.0·10 <sup>-9</sup> (f); 7.8·10 <sup>-10</sup> (s)	5.0·10 <sup>-9</sup> (f); 7.8·10 <sup>-10</sup> (s)	5.0·10 <sup>-9</sup> (f); 7.8·10 <sup>-10</sup> (s)	3.1·10 <sup>-9</sup>	1.2·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	1.9·10 <sup>-8</sup>	1.5·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	1.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>												6-9
molecular weight	g.mol <sup>-1</sup>	148.1	406.92	380.93	242.3	277.25	409	385.5	367	278.34	296.56	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	138	3980	1.6·10 <sup>5</sup>	3900	2510	1905	12589	1905	12600	4300	
melting point	°C	86	83.7	209	liquid	liquid	122	166	119	liquid	177	
vapor pressure (25 °C)	Pa	1.1·10 <sup>-2</sup>	1.3·10 <sup>-3</sup>	2.0·10 <sup>-5</sup>	5.1·10 <sup>-2</sup>	1.3·10 <sup>-4</sup>	2.3·10 <sup>-4</sup>	1	9.9·10 <sup>-6</sup>	4.0·10 <sup>-3</sup>	1.3·10 <sup>-3</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	150 <sup>b</sup>	0.5	0.23	750	30	9.6	76	1.1	50	1	
dissociation constant	-	4.2										
<b>Partition coefficients</b>												6, 7
organic carbon partition coefficient	l.kg <sup>-1</sup>	219	12300	19500	68	1350	21900	21900	21900	1510	1900	
<b>Degradation rates</b>												7, 9-15
reaction half-life in air	d	77.5	6.2·10 <sup>-1</sup>	6.0·10 <sup>-2</sup>								
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>				6.7·10 <sup>-11</sup>	6.2·10 <sup>-11</sup>				7.6·10 <sup>-11</sup>	1.4·10 <sup>-11</sup>	
hydrolysis in water (PH=6, 12 °C)	d		21			147					351	
hydrolysis in water (PH=7, 12 °C)	d		15			147					349	
hydrolysis in water (PH=8, 12 °C)	d		3.5			145					338	
biodegradation in water (12 °C)	d	21	9	4871	161	22	12.21	12.21	12.21	30	1000	
hydrolysis in soil (PH=6 12 °C)	d		210			147					351	
hydrolysis in soil, sediment (PH=7, 12 °C)	d		150			147					349	
hydrolysis in sediment (PH=8, 12 °C)	d		35			145					338	
biodegradation in soil (12 °C)	d	21	9	4871	56	49	244	244	244	59	1000	
aerobic biodegradation in sediment (12 °C)	d	21	9	4871	56	49	2442	2442	2442	59	1000	
anaerobic biodegradation in sediment (12 °C)	d	8	37	15	223	195	9766	9766	9766	237	4000	
metabolism in plant tissue	d					4						
<b>Exposure assessment</b>												7, 16-22
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	0.2	601	1873		152	1095	1095	1095	759		
transpiration stream concentration factor	-	0.04 <sup>c</sup>										
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>	0.7 <sup>c</sup>		12								
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>			1.7·10 <sup>-2</sup>								
biotransfer factor for meat	d.kg <sup>-1</sup>		7.2·10 <sup>-4</sup>	1.4·10 <sup>-2</sup>						2.4·10 <sup>-5</sup>		
biotransfer factor for milk	d.kg <sup>-1</sup>			2.1·10 <sup>-3</sup>		7.4·10 <sup>-6</sup>				6.7·10 <sup>-5</sup>		

bioavailability for oral uptake	-						0.4	0.4	0.4		
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<sup>a</sup>EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>c</sup> assumed default value for dissociating acids at environmental pH of 7, based on Briggs et al. (1987); f = fresh water; s = salt water; <sup>1</sup> WHO/FAO (1998); <sup>2</sup> FAO/WHO (1999); <sup>3</sup> Environmental Defense Fund (1999); <sup>4</sup> RIZA (1999); <sup>5</sup> Huijbregts (1999); <sup>6</sup> Crommentuijn et al. (1997b); <sup>7</sup> Mackay et al. (1997); <sup>8</sup> Guinée et al. (1996a); <sup>9</sup> Tomlin (1994); <sup>10</sup> Howard et al. (1991); <sup>11</sup> RIVM et al. (1998); <sup>12</sup> Syracuse Research Corporation (1993); <sup>13</sup> Linders et al. (1994); <sup>14</sup> Jager et al. (1997); <sup>15</sup> Van Rijn et al. (1995); <sup>16</sup> Slooff et al. (1993); <sup>17</sup> Nendza (1991); <sup>18</sup> Tsuda et al. (1997); <sup>19</sup> Polder et al. (1995); <sup>20</sup> Dowdy & McKone (1997); <sup>21</sup> Dowdy et al. (1996); <sup>22</sup> Van de Plassche (1994).

Table B.8f: Substance-specific input parameters for pesticides

Compound name	Unit	Glyphosate	Heptachlor	Heptenophos	Iprodione	Isoproturon	Lindane	Linuron	Malathion	MCPA	Mecoprop	Source
CAS nr.	-	1071-83-6	76-44-8	23560-59-0	36734-19-7	34123-59-6	58-89-9	330-55-2	121-75-5	94-74-6	7085-19-0	
<b>Effects assessment</b>												1-8
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.0·10 <sup>-7</sup>	1.0·10 <sup>-10</sup>	5.0·10 <sup>-9</sup>	6.0·10 <sup>-8</sup>	6.2·10 <sup>-9</sup>	1.0·10 <sup>-9</sup>	2.0·10 <sup>-9</sup>	3.0·10 <sup>-7</sup>	1.5·10 <sup>-9</sup>	1.0·10 <sup>-9</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>						2.5·10 <sup>-10</sup>			7.0·10 <sup>-9</sup>		
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.6·10 <sup>-6</sup>	8.6·10 <sup>-9</sup>	2.0·10 <sup>-8</sup>	2.3·10 <sup>-6</sup>	3.2·10 <sup>-6</sup>	1.0·10 <sup>-6</sup>	2.5·10 <sup>-7</sup>	1.3·10 <sup>-8</sup>	4.2·10 <sup>-5</sup>	3.9·10 <sup>-6</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>a</sup>	2.0·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	2.5·10 <sup>-9</sup>	EP <sup>a</sup>	5.0·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>												5, 9, 10
molecular weight	g.mol <sup>-1</sup>	169.1	373.4	250.6	330.2	206.3	290.85	249.1	330.36	200.6	214.6	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	3.1·10 <sup>-2</sup>	1.9·10 <sup>5</sup>	209	1010	178	5010	1010	631	490	8710	
melting point	°C	200	95.5	liquid	134	155.5	112.5	93.5	2.9	118.5	94.5	
vapor pressure (25 °C)	Pa	4.0·10 <sup>-5</sup>	5.3·10 <sup>-2</sup>	1.7·10 <sup>-1</sup>	5.0·10 <sup>-7</sup>	3.3·10 <sup>-6</sup>	3.7·10 <sup>-3</sup>	2.3·10 <sup>-2</sup>	1.0·10 <sup>-3</sup>	2.0·10 <sup>-4</sup>	3.1·10 <sup>-4</sup>	
solubility (25 °C)	mg.L <sup>-1</sup>	<sup>b</sup> 1.3·10 <sup>5</sup>	6.0·10 <sup>-2</sup>	2360	14	55	7.3	75	145	1605 <sup>b</sup>	620 <sup>b</sup>	
dissociation constant	-	5.7								3.1	3.7	
<b>Partition coefficients</b>												9-11
organic carbon partition coefficient	l.kg <sup>-1</sup>	3630	24000			72	955	603	1170	54	8	
<b>Degradation rates</b>												5, 10, 12-16
reaction half-life in air	d		2.5·10 <sup>-1</sup>			1.9	2.31	1.2·10 <sup>-1</sup>	2.5·10 <sup>-1</sup>	1.3		
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	7.5·10 <sup>-11</sup>		3.0·10 <sup>-11</sup>	4.6·10 <sup>-11</sup>							1.7·10 <sup>-11</sup>
hydrolysis in water (PH= 6, 12 °C)	d	26	4		11		413					
hydrolysis in water (PH= 7, 12 °C)	d				5		361					
hydrolysis in water (PH= 8, 12 °C)	d				1		159					
biodegradation in water (12 °C)	d	105	54	4	8	74	197	123	26	9	12	
hydrolysis in soil (PH=6, 12 °C)	d	26	41		11		413					
hydrolysis in soil (PH=7, 12 °C)	d				5		361					
biodegradation in soil (12C)	d	105	54	1	72	74	197	123	8	9	12	
hydrolysis in sediment (PH=7, 12 °C)	d	261	405		5		361					
hydrolysis in sediment (PH=8, 12 °C)	d				1		159					
aerobic biodegradation in sediment (12 °C)	d	1046	54	1	72	74	197	123	26	9	12	
anaerobic biodegradation in sediment (12 °C)	d	4186	218	5	286	14	23	493	103	124	49	
metabolism in plant tissue	d	7.9					9.5	8.1	0.6			
<b>Exposure assessment</b>												10, 17-23
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>		7427				371		64			
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>						3.3·10 <sup>5</sup>					
transpiration stream concentration factor	-	0.04 <sup>c</sup>			0.79			0.93		0.04 <sup>c</sup>	0.04 <sup>c</sup>	
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>	0.7 <sup>c</sup>					17			0.7 <sup>c</sup>	0.7 <sup>c</sup>	
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>		9.8·10 <sup>-3</sup>				6.3·10 <sup>-2</sup>					

biotransfer factor for meat	d.kg <sup>-1</sup>		1.3·10 <sup>-2</sup>				1.5·10 <sup>-2</sup>		9.1·10 <sup>-6</sup>		
biotransfer factor for milk	d.kg <sup>-1</sup>		1.4·10 <sup>-3</sup>				2.2·10 <sup>-3</sup>		1.7·10 <sup>-6</sup>	1.1·10 <sup>-5</sup>	
bioavailability for oral uptake	-	0.3									

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> solubility of the neutral species; <sup>c</sup> assumed default value for dissociating acids at environmental pH of 7, based on Briggs et al. (1987); <sup>1</sup> FAO/WHO (1998); <sup>2</sup> Janus et al. (1994); <sup>3</sup> RIZA (1999); <sup>4</sup> Huijbregts (1999); <sup>5</sup> Tomlin (1994); <sup>6</sup> Janssen et al. (1998); <sup>7</sup> Environmental Defense Fund (1999); <sup>8</sup> USEPA (1998b); <sup>9</sup> Crommentuijn et al. (1997b); <sup>10</sup> Mackay et al. (1997); <sup>11</sup> Sabljic et al. (1995); <sup>12</sup> Howard et al. (1991); <sup>13</sup> RIVM et al. (1998); <sup>14</sup> Syracuse Research Corporation (1993); <sup>15</sup> Linders et al. (1994); <sup>16</sup> Komoša et al. (1995); <sup>17</sup> Sicbaldi et al. (1997); <sup>18</sup> Polder et al. (1995); <sup>19</sup> Polder et al. (1998); <sup>20</sup> Dowdy & McKone (1997); <sup>21</sup> Dowdy et al. (1996); <sup>22</sup> WHO (1994a); <sup>23</sup> Van de Plassche (1994)

Table B.8g: Substance-specific input parameters for pesticides

Compound name	Unit	Metamitron	Metazachlor	Methabenz-thiazuron	Methomyl	Methyl-bromide	Meto-bromuron	Metolachlor	Mevinphos	Oxamyl	Oxydemethon-methyl	Source
CAS nr.	-	41394-05-2	67129-08-2	18691-97-9	16752-77-5	74-83-9	3060-89-7	51218-45-2	7786-34-7	23135-22-0	301-12-2	
<b>Effects assessment</b>												1-5
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	1.3·10 <sup>-7</sup>	3.6·10 <sup>-8</sup>	5.0·10 <sup>-8</sup>	3.0·10 <sup>-8</sup>	1.0·10 <sup>-6</sup>	<sup>a</sup> 3.0·10 <sup>-8</sup>	1.5·10 <sup>-7</sup>	8.0·10 <sup>-10</sup>	3.0·10 <sup>-8</sup>	3.0·10 <sup>-10</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.0·10 <sup>-4</sup>	3.4·10 <sup>-5</sup>	8.4·10 <sup>-6</sup>	8.0·10 <sup>-8</sup>	1.1·10 <sup>-5</sup>	3.6·10 <sup>-5</sup>	2.0·10 <sup>-7</sup>	1.6·10 <sup>-9</sup>	1.8·10 <sup>-6</sup>	3.5·10 <sup>-8</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	8.3·10 <sup>-8</sup>	EP <sup>b</sup>	EP <sup>b</sup>	EP <sup>b</sup>	
<b>Physico-chemical properties</b>												3, 6-9
molecular weight	g.mol <sup>-1</sup>	202.2	277.8	221.3	162.2	94.94	259.1	283.8	224.1	219.25	246.3	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	5	100	204	4	15	257	1350	3	0.4	0.2	
melting point	°C	166.6	85	120	78.5	-93	95.75	liquid	-56.1	101	-20	
vapor pressure (25 °C)	Pa	2.0·10 <sup>-6</sup>	6.9·10 <sup>-5</sup>	1.5·10 <sup>-5</sup>	6.7·10 <sup>-3</sup>	1700	4.0·10 <sup>-4</sup>	4.2·10 <sup>-3</sup>	1.7·10 <sup>-2</sup>	3.1·10 <sup>-2</sup>	5.4·10 <sup>-3</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	1820	461	63	58000	14000	330	430	6.0·10 <sup>-5</sup>	2.8·10 <sup>5</sup>	25000	
<b>Partition coefficients</b>												6, 7, 10
Henry's law constant (25 °C)	Pa.m <sup>3</sup> .mol <sup>-1</sup>					630						
organic carbon partition coefficient	1.kg <sup>-1</sup>	158	129	631	23		186	214	631	11		
<b>Degradation rates</b>												3, 7, 9, 11-17
reaction half-life in air	d					408				1.5·10 <sup>-1</sup>		
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	3.7·10 <sup>-11</sup>	6.7·10 <sup>-11</sup>	3.2·10 <sup>11</sup>	1.6·10 <sup>-11</sup>		3.7·10 <sup>-11</sup>	7.0·10 <sup>-11</sup>		2.9·10 <sup>-11</sup>	1.1·10 <sup>-10</sup>	
hydrolysis in water, soil, sediment (PH=6, 12 °C)	d	54			457	23			209	33	99	
hydrolysis in water, soil sediment (PH=7, 12 °C)	d	54			457	21			61	14	80	
hydrolysis in water, soil, sediment (PH=8, 12 °C)	d	45			457	16			30	2	26	
biodegradation in water (12 °C)	d	33	58	157	400	24	1000	105	9	30	30	
biodegradation in soil (12 °C)	d	52	31	235	52	24	1000	105	4	17	1	
aerobic biodegradation in sediment (12 °C)	d	52	31	235	52	24	1000	105	4	17	1	
anaerobic biodegradation in sediment (12 °C)	d	209	126	942	209	98	4000	419	14	69	4	
metabolism in plant tissue	d				4				1	1.8		
<b>Exposure assessment</b>												7, 9, 17
bioconcentration factor in fish	1.kg(wwt) <sup>-1</sup>							7				
bioconcentration factor from porewater to roots	1.kg(wwt) <sup>-1</sup>									0.9		
bioavailability for oral uptake	-					5.5·10 <sup>-1</sup>						

<sup>a</sup> The oral Human Limit Values of Metobromuron is derived by dividing a NOEL for dogs (2 years), listed in Tomlin (1994), by a factor 100 (intra- and interspecies extrapolation) (personal assessment); <sup>b</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> FAO/WHO (1998); <sup>2</sup> RIZA (1999); <sup>3</sup> Tomlin (1994); <sup>4</sup> Huijbregts (1999); <sup>5</sup> USEPA (1998b); <sup>6</sup> Crommentuijn et al. (1997b); <sup>7</sup> Mackay et al. (1997); <sup>8</sup> Camergesoft Corporation (1998); <sup>9</sup> WHO (1995b); <sup>10</sup> Lide (1993); <sup>11</sup> Howard et al. (1991); <sup>12</sup> RIVM et al. (1998); <sup>13</sup> Syracuse Research Corporation (1993); <sup>14</sup> Jager et al. (1997); <sup>15</sup> Linders et al. (1994); <sup>16</sup> Van Rijn et al. (1995); <sup>17</sup> Briggs et al. (1982)

Table B.8h: Substance-specific input parameters for pesticides

Compound name	Unit	Parathion-ethyl	Parathion-methyl	Permethrin	Phoxim	Pirimicarb	Propachlor	Propoxur	Pyrazophos	Simazine	Source
CAS nr.	-	56-38-2	298-00-0	52645-53-1	14816-18-3	23103-98-2	1918-16-7	114-26-1	13457-18-6	122-34-9	
<b>Effects assessment</b>											1-4
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	4.0·10 <sup>-9</sup>	3.0·10 <sup>-9</sup>	5.0·10 <sup>-8</sup>	1.0·10 <sup>-9</sup>	2.0·10 <sup>-8</sup>	1.3·10 <sup>-8</sup>	2.0·10 <sup>-8</sup>	4.0·10 <sup>-9</sup>	5.0·10 <sup>-9</sup>	
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.9·10 <sup>-9</sup>	1.1·10 <sup>-8</sup>	3.0·10 <sup>-10</sup>	8.2·10 <sup>-8</sup>	8.2·10 <sup>-8</sup>	1.3·10 <sup>-6</sup>	1.0·10 <sup>-8</sup>	4.0·10 <sup>-8</sup>	1.4·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	1.0·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	1.4·10 <sup>-9</sup>	
<b>Physico-chemical properties</b>											5-7
molecular weight	g.mol <sup>-1</sup>	291.27	263.5	391.3	298.3	238.3	211.7	209.24	373.4	201.7	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	6310	1010	1.3·10 <sup>6</sup>	2400	50	151	32	6300	151	
melting point	°C	6	37.5	36.4	6.1	90.5	71.4	91.5	51.5	226	
vapor pressure (25 °C)	Pa	6.0·10 <sup>-4</sup>	2.0·10 <sup>-3</sup>	1.7·10 <sup>-6</sup>	3.0·10 <sup>-3</sup>	9.7·10 <sup>-4</sup>	3.0·10 <sup>-2</sup>	1.7·10 <sup>-5</sup>	4.6·10 <sup>-5</sup>	8.5·10 <sup>-6</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	12.4	25	1.0·10 <sup>-2</sup>	1.6·10 <sup>-3</sup>	3210	600	1800	4.2	5	
<b>Partition coefficients</b>											5, 7
organic carbon partition coefficient	l.kg <sup>-1</sup>	10500	1480	24500		417	79	22		110	
<b>Degradation rates</b>											5, 6, 8-12
reaction half-life in air	d		2.6·10 <sup>-1</sup>					1.8·10 <sup>-1</sup>			
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>	9.0·10 <sup>-11</sup>		3.9·10 <sup>-11</sup>	9.8·10 <sup>-11</sup>	2.3·10 <sup>-10</sup>	2.2·10 <sup>-11</sup>		1.1·10 <sup>-10</sup>	1.4·10 <sup>-10</sup>	
hydrolysis in water (PH=6, 12 °C)	d	112	94		12			2964			
hydrolysis in water (PH=7, 12 °C)	d	328	94		13			372			
hydrolysis in water (PH=8, 12 °C)	d	183	89		10			30			
biodegradation in water (12 °C)	d	37	57	35	1000	30	14	24	17	37	
hydrolysis in soil (PH=6, 12 °C)	d	1120	94		12			2964			
hydrolysis in soil (PH=7, 12 °C)	d	3280	94		13			372			
biodegradation in soil (12C)	d	37	105	35	1000	188	11	24	68	105	
hydrolysis in sediment (PH=7, 12 °C)	d	3280	936		13			372			
hydrolysis in sediment (PH=8, 12 °C)	d	1830	889		10			30			
aerobic biodegradation in sediment (12 °C)	d	37	57	35	1000	188	11	24	68	105	
anaerobic biodegradation in sediment (12 °C)	d	147	5	138	4000	753	43	98	272	420	
metabolism in plant tissue	d	0.7									
photodegradation upon plant tissue	d	20.3					1.6				
<b>Exposure assessment</b>											5, 13-19
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	159	302	2522							
transpiration stream concentration factor	-									0.9	
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>									4.5	
bioconcentration factor from soil to leafs	kg(wwt).kg(wwt) <sup>-1</sup>									0.2	

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>1</sup> FAO/WHO (1998); <sup>2</sup> RIZA (1999); <sup>3</sup> Huijbregts (1999); <sup>4</sup> USEPA (1998b); <sup>5</sup> Mackay et al. (1997); <sup>6</sup> Tomlin (1994); <sup>7</sup> Crommentuijn et al. (1997b); <sup>8</sup> Howard et al. (1991); <sup>9</sup> RIVM et al. (1998); <sup>10</sup> Syracuse Research Corporation (1993); <sup>11</sup> Jager et al. (1997); <sup>12</sup> Linders et al. (1994); <sup>13</sup> Schwack et al. (1994); <sup>14</sup> Schynowski & Schwack (1996); <sup>15</sup> Cabras et al. (1990); <sup>16</sup> Komoša et al. (1995); <sup>17</sup> Sicbaldi et al. (1997); <sup>18</sup> Polder et al. (1995); <sup>19</sup> Dowdy & McKone (1997)

Table B.8i: Substance-specific input parameters for pesticides

Compound name	Unit	2,4,5-T	Thiram	Tolclophos-methyl	Tri-allaat	Triazophos	Tributyltin-oxide	Trichlorfon	Trifluralin	Zineb	Source
CAS nr.	-	93-76-5	137-26-8	57018-04-9	2303-17-5	24017-47-8	56-35-9	52-68-6	1582-09-8	12122-67-7	
<b>Effects assessment</b>											1-6
Oral Human Limit Value	kg.kg(bw) <sup>-1</sup> .d <sup>-1</sup>	3.0·10 <sup>-8</sup>	1.0·10 <sup>-8</sup>	7.0·10 <sup>-8</sup>	1.3·10 <sup>-8</sup>	1.0·10 <sup>-9</sup>	3.0·10 <sup>-10</sup>	1.0·10 <sup>-8</sup>	7.5·10 <sup>-9</sup>	3.0·10 <sup>-8</sup>	
Inhalatory Human Limit Value	kg.m <sup>-3</sup>						2.0·10 <sup>-11</sup>				
Aquatic Predicted No Effect Concentration	kg.m <sup>-3</sup>	1.6·10 <sup>-4</sup>	3.2·10 <sup>-8</sup>	7.9·10 <sup>-7</sup>	8.0·10 <sup>-8</sup>	3.2·10 <sup>-8</sup>	1.4·10 <sup>-8</sup> (f); 1.0·10 <sup>-9</sup> (s)	1.0·10 <sup>-9</sup>	2.6·10 <sup>-8</sup>	2.0·10 <sup>-7</sup>	
Terrestrial Predicted No Effect Concentration	kg.kg(dwt) <sup>-1</sup>	1.5·10 <sup>-8</sup>	EP <sup>a</sup>	EP <sup>a</sup>	4.4·10 <sup>-9</sup>	EP <sup>a</sup>	EP <sup>a</sup>	EP <sup>a</sup>	1.4·10 <sup>-9</sup>	EP <sup>a</sup>	
<b>Physico-chemical properties</b>											7-11
molecular weight	g.mol <sup>-1</sup>	255.5	240.4	301.1	304.7	313.3	596	257.45	335.5	275.8	
octanol-water partition coefficient	m <sup>3</sup> .m <sup>-3</sup>	1350	54	36000	19500	2794	6310	3	2.2·10 <sup>5</sup>	20	
melting point	°C	153	145	79	29.5	3.2	-45	83.5	48.75	157	
vapor pressure (25 °C)	Pa	5.0·10 <sup>-3</sup>	1.3·10 <sup>-3</sup>	5.7·10 <sup>-2</sup>	1.5·10 <sup>-2</sup>	2.8·10 <sup>-4</sup>	1.4·10 <sup>-3</sup>	1.0·10 <sup>-3</sup>	2.6·10 <sup>-2</sup>	1.3·10 <sup>-5</sup>	
solubility (25 °C)	mg.l <sup>-1</sup>	220 <sup>c</sup>	30	0.3	4	37.1	71.2	154000	0.5	10	
dissociation constant	-	2.9									
<b>Partition coefficients</b>											7, 8
organic carbon partition coefficient	l.kg <sup>-1</sup>	98	490	2000	1380	355	12600	38	8510	400	
<b>Degradation rates</b>											8-10, 12-19
reaction half-life in air	d	3.1	26.6					2.5			
hydroxyl radical reaction in air	cm <sup>3</sup> .molec <sup>-1</sup> .s <sup>-1</sup>			6.3·10 <sup>-11</sup>	3.2·10 <sup>-11</sup>	1.1·10 <sup>-10</sup>			8.4·10 <sup>-11</sup>		
hydrolysis in water, soil, sediment (PH=6, 12 °C)	d		56					8			
hydrolysis in water, soil sediment (PH=7, 12 °C)	d		31					4			
hydrolysis in water, soil, sediment (PH=8, 12 °C)	d		5					2			
biodegradation in water (12 °C)	d	25	1000	53	105	61	224	12	105	63	
biodegradation in soil (12 °C)	d	25	1000	115	12	113	244	12	105	49	
aerobic biodegradation in sediment (12 °C)	d	25	1000	115	12	113	244	12	105	49	
anaerobic biodegradation in sediment (12 °C)	d	124	4000	460	47	453	976	47	49	195	
metabolism in plant tissue	d	7.8									
<b>Exposure assessment</b>											8, 10, 20-24
bioconcentration factor in fish	l.kg(wwt) <sup>-1</sup>	43		579			510		3467		
partitioning coefficient between leaves and air	m <sup>3</sup> .m <sup>-3</sup>								1.0·10 <sup>5</sup>		
plant conductance	m.s <sup>-1</sup>	6.39·10 <sup>-4</sup>									
transpiration stream concentration factor	-	0.04 <sup>d</sup>									
bioconcentration factor from porewater to roots	l.kg(wwt) <sup>-1</sup>	0.7 <sup>d</sup>									
biotransfer factor for meat	d.kg <sup>-1</sup>	2.0·10 <sup>-5</sup>							9.4·10 <sup>-7</sup>		
biotransfer factor for milk	d.kg <sup>-1</sup>	2.0·10 <sup>-5</sup>									
bioavailability for oral uptake	-						3.5·10 <sup>-1</sup>				

<sup>a</sup> EP = PNEC<sub>soil</sub> derived by Equilibrium Partitioning; <sup>b</sup> The fresh water PNEC is used in the derivation of the soil PNEC; <sup>c</sup> solubility of the neutral species; <sup>d</sup> default value for dissociating acids at environmental pH of 7, based on Briggs et al. (1987); f = fresh water; s = salt water; <sup>1</sup> FAO/WHO (1998); <sup>2</sup> RIZA (1999); <sup>3</sup> USEPA (1998b); <sup>4</sup> Janssen et al. (1995); <sup>5</sup> Huijbregts (1999); <sup>6</sup>

Environmental Defense Fund (1999);<sup>7</sup> Crommentuijn et al. (1997b);<sup>8</sup> Mackay et al. (1997);<sup>9</sup> Tomlin (1994);<sup>10</sup> Slooff et al. (1993);<sup>11</sup> Guinée et al. (1996a);<sup>12</sup> Howard et al. (1991);<sup>13</sup> RIVM et al. (1998);<sup>14</sup> Howard (1991);<sup>15</sup> Syracuse Research Corporation (1993);<sup>16</sup> Jager et al. (1997);<sup>17</sup> Linders et al. (1994);<sup>18</sup> Van Rijn et al. (1995);<sup>19</sup> Komôša et al. (1995);<sup>20</sup> Paterson et al. (1991);<sup>21</sup> Tsuda et al. (1997);<sup>22</sup> Dowdy et al. (1996);<sup>23</sup> Garten & Trabalka (1983);<sup>24</sup> Riederer (1995)